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Epitaxial growth of tantalum carbides by low carbon flow carburizing

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H I G H L I G H T S

- Different kinds of carbides could appear on a tantalum sample if the carbon flux varies on its surface.
- Epitaxial relationships between Ta, Ta₂C and TaC are highlighted in tantalum carbide layers.
- Under specifics carbon flux, TaC carbide layer grow in relation with the Ta substrate.
- Columnar and equiaxed growths of carbide layers are separated by a carbon boundary flux.

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Keywords:

Carbides
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A B S T R A C T

The carburizing of tantalum samples under different carbon flow rates has highlighted the influence of the carbon flow rate on the structure of tantalum carbides. XRD analyses enabled the identification of surface structures that were revealed by optical micrographs. Four carbon flow rates were tested. The lowest carbon flow rate produced a Ta₂C layer that grew epitaxially on the tantalum substrate according to the relationship: $\{10\bar{1}\}_{Ta} // \{\bar{1}101\}_{Ta_2C}$; $\langle 10\bar{1} \rangle_{Ta} // \langle \bar{1}717010 \rangle_{Ta_2C}$. An increase in the carbon flow rate resulted in the appearance of TaC nuclei through the carbon enrichment of the Ta₂C layer. These nuclei appeared in the form of islets and highly oriented needles. EBSD (Electron Back Scattered Diffraction) analyses and pole figures revealed that the carbide layers grew epitaxially towards their substrates. The Ta₂C layer grew according to the relationship $\{10\bar{1}\}_{Ta} // \{\bar{1}101\}_{Ta_2C}$ while the TaC structures nucleated according to the following relationships: $\{111\}_{TaC} // \{0001\}_{Ta_2C}$; $\langle 111 \rangle_{TaC} // \langle 0001 \rangle_{Ta_2C}$; $\{100\}_{TaC} // \{20\bar{2}3\}_{Ta_2C}$; $\langle 100 \rangle_{TaC} // \langle 90\bar{9}8 \rangle_{Ta_2C}$ and $\{110\}_{TaC} // \{10\bar{1}3\}_{Ta_2C}$; $\langle 110 \rangle_{TaC} // \langle 90\bar{9}16 \rangle_{Ta_2C}$. The highest carbon flow produced stoichiometric TaC grains at the surface, which appeared to be equiaxed. The diversity of the TaC crystallographic orientations stems from the underlying Ta₂C layer, the TaC grains having nucleated epitaxially from the Ta₂C layer. This misorientation of the Ta₂C grains with respect to Ta is governed by the exceeding of a carbon flow limit.

1. Introduction

Carburizing process increases the concentration of carbon at the surface of a metallic part. This increase in carbon concentration improves certain surface properties for a variety of applications: an increase in hardness for better contact conditions, the creation of a thermal barrier, the creation of a catalytic layer, etc. For example,

the tantalum carbide TaC is a compound that can be produced with a carburizing process. It is a compound of extreme hardness, which can exceed 5000 HK [1]. This property makes it the coating material of choice for cutting tool inserts [2]. This material has also been extensively studied for its refractory properties [3,4] on account of its high melting temperature of nearly 4270 K [5,6].

Some carburizing techniques for tantalum are described in the literature. The oldest procedure for tantalum carburizing uses solid carbon [7,8], with temperatures higher than 2000 °C. Other procedures use hydrocarbons as a carburizing gas, such as atmospheric carburizing [9] or low-pressure carburizing. The low pressure carburizing consist of injecting a carburizing gas such as CO, CO₂ or a hydrocarbon (CH₄, C₂H₂, C₂H₄, C₃H₈, ...) into the chamber

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[10–13]. This type of procedure can be complemented by the use of plasma [14,15].

Tantalum carburizing gives rise to the appearance of two main carbides: TaC and Ta₂C [16]. TaC is an FCC (Face Centered Cubic) compound (NaCl- or B1-type structure) [6,16]. According to Lesser et al. [17], TaC can exist in a non-stoichiometric form, and can contain between 42.5 and 50 at.% carbon. This variation in stoichiometry brings about changes in the lattice parameter [5,12,17–19], which can vary between 4.41 and 4.4545 Å. Ta₂C has two allotropic phases, α -Ta₂C (ordered carbon sublattice, CdI₂ antitype structure) and β -Ta₂C (disordered carbon sublattice, L'3 structure) [6,20]. The allotropic transformation between α - and β -Ta₂C occurs at approximately 1950 K [21].

One of the more recent phase diagrams [21] describes two other tantalum carbides: Ta₄C₃ (ζ phase) and Ta₆C₅. Ta₄C₃ was described for the first time by Lesser et al. [17]. It is a rhombohedral carbide (R3m space group) that appears in the presence of TaC and Ta₂C [17,22,23]. Ta₆C₅ is an ordered phase created by the order–disorder transition of non-stoichiometric TaC_x [24,25]. These two carbides appear after long annealing times.

Many papers deal with the growth of tantalum carbide layers [7,22,26–31]. These papers describe carbide layers with thicknesses ranging from a few tens of microns to several hundred microns. These layers always have a two-phase structure, with TaC at the surface and Ta₂C underneath [26,28,29,32]. Other papers [7,27,30,31] report a three-phase structure with Ta₄C₃ interposed between TaC and Ta₂C layers.

These layers were analyzed based on micrographs, XRD and XPS in order to identify the nature of the phases [32]. However, the crystallographic orientations of carbide layers produced by carburizing have not been addressed in the literature. Very few studies have dealt with the mechanisms of the development of tantalum carbide layers through tantalum carburizing. Based on the results obtained so far, the following mechanism has been proposed [32]:

- The first carbide layer that forms is Ta₂C. It is produced by the saturation of tantalum with carbon.
- Next, a layer of TaC develops through carbon enrichment of the Ta₂C layer.

These hypotheses are based on micrographic observations of two-phase TaC+Ta₂C structures. Currently, no experimental confirmation of such a formation mechanism has been reported in the literature. The present study was conducted with the aim of observing the very first stages in the development of tantalum carbide layers. The nature of the carbide layers was determined by XRD. The crystallographic orientations of the various phases were studied by EBSD.

2. Experimental procedure

2.1. Materials and tantalum carburizing

During carburizing, tantalum becomes saturated with carbon very quickly, which results in the appearance and rapid growth of surface carbide layers [32]. To observe the very first stages in the development of these layers, the extent to which tantalum becomes saturated with carbon at the surface must be reduced. There are two ways of achieving this: increase the solubility of carbon in tantalum by increasing the temperature or decrease the carbon supply from carburizing.

According to the Ta–C phase diagram [21], the solubility of tantalum can be increased to 7.5 at.% by raising the carburizing temperature to 3015 K. However, various technical problems prevent this temperature from being reached. Thus, limiting the

carbon supply appears to be the easier option.

Samples measuring 5 mm thick and 14 mm in diameter were machined from a bar of tantalum (99.9% purity). Carburizing tests were carried out in a thermochemical treatment furnace under reduced pressure. The carburizing procedure comprised of the following steps: creation of a vacuum in the chamber, heating, carburizing itself and finally, cooling. The pressure was reduced to less than 0.1 mbar. The carburizing temperature (1600 °C) was attained with a heating rate of 30 °C/min. The tantalum samples were then carburized by injecting 20 mbar of hydrogen into the chamber containing solid carbon. The gas was kept in the chamber for 5 min, after which a primary vacuum was created and maintained for one hour. The samples were then cooled in N₂ at a pressure of 1 bar.

The above procedure was chosen with the aim of reducing the carbon supply during carburizing. During time when the chamber was filled with hydrogen, the gas reacted with the solid carbon to form hydrocarbons. The vacuum created drastically reduced the partial pressure of the hydrocarbons, thereby reducing the rate of carburizing of tantalum.

The holder and the resistors are made of graphite. The holder is a possible source of carbon, but at 1600 °C, the carbon vapor pressure is too low to carburize tantalum [29]. A media gas, such as hydrogen, is needed to carry the carbon from the holder to the tantalum sample. Moreover, the holder side carburizing is not homogeneous. The carburizing is higher on the edge than in the centre. If the holder would have been a carbon source, the carburizing would have been homogeneous.

According to the bibliography, the carbon source is hydrocarbons which result from the reaction between hydrogen and graphite parts [33]. Between 1400 °C and 1600 °C, thermodynamic reaction data between carbon and hydrogen [34,35] show that the most abundant gases produced are methane, acetylene and ethylene, according to reactions (1) to (3). The production of the methyl radical CH₃• is also noted (4).



These gases are known to carburize tantalum at temperatures used here [11,14,15,36]. The thermodynamic data are given for gas in equilibrium with carbon. This is not the case of carburizing, because the produced hydrocarbon gases gradually disappear by tantalum carburizing.

Hydrocarbons react with tantalum surface and let adsorbed carbon and gaseous hydrogen. The adsorbed carbon diffuses in tantalum. Hydrogen could react with graphite's holder to create hydrocarbons again. This reaction would lead to a homogeneous carburizing of the holder surface but this is not the case. Carbon might come from another mechanism. The carburizing reaction between hydrogen and graphite parts is not yet well understood in this furnace.

2.2. Surface preparation

The tantalum samples possessed two flat parallel faces. Before carburizing, these faces were mechanically polished. The final step is achieved with a 3- μm diamond paste. During carburizing, the samples were placed on a graphite holder. For the remainder of this paper, the free face will be referred to as the 'chamber side' while

the face that was in contact with the holder will be called the 'holder side'.

As the chamber side encountered a strong carbon flow, this led to the formation of tantalum carbide layers identical to those described in the literature. The carburizing technique that was used did not sufficiently reduce the carbon flow on the chamber side. Observing the very first instants of carbide layer formation requires a much lower carbon flow than that experienced by the chamber side. In addition to a special carburizing technique, it was necessary to introduce a means of containment to limit the carbonaceous gases' access to the tantalum and reduce the carbon flow effectively.

The surface roughness of the holder meant that there was a confined space between the sample and the holder. This provided us with an effective means of limiting the carbon flow rate. The structures that formed on the holder side enabled an investigation of the initial stages of the development of tantalum carbide layers. Therefore, only this side was studied. The microstructures showed in this paper were observed on a representative sample.

The structures produced by carburizing were left as is, they were not chemically etched. The samples were cross-sectioned in order to study these structures across the depth. These cross-sections were polished until a finishing step involving the use of colloidal silica. The cross-sections that were intended for EBSD analysis required an additional step (vibration polishing for 12 h).

2.3. Analysis of surface microstructures

A series of optical micrographs was obtained for several locations from the centre to the edge of the surface. Top views of the structures were examined. An Olympus BX51M microscope was used for these analyses.

The cross-sections were studied by means of a Field Emission SEM (JEOL JSM 7600-F). The accelerating voltage of the electron beam was 20 kV. The crystallographic orientations of the surface structures and the tantalum substrate were studied through EBSD observations of the cross-sections, with the help of a TSL EDAX OIM XM detector coupled to the Field Emission SEM. The step size (distance between each point that was analyzed) was 25 nm.

The microstructures that developed at the surface were identified by MicroXRD, with a Bruker D8 Discover device. This device is equipped with a cobalt anticathode ($\lambda_{K\alpha 1} = 1.7889 \text{ \AA}$) and a collimator that reduces the beam diameter from 2000 to 50 μm . The areas that were analyzed in this study measured 500 μm in diameter.

2.4. Processing of EBSD Data

Ta₂C has a trigonal L/3 structure. During EBSD analysis, this type of structure can result in ambiguous plane indexing because of its pseudosymmetry [37,38]. This is reflected in the raw crystallographic orientation data, which sometimes indicate two different orientations within the same grain (Fig. 1a and b). Such erroneous indexing can be corrected by using an appropriate technique to process the raw data. The pseudosymmetry relationship must be incorporated in the processing software. In this study, this relationship was determined as 60° in the [0001] direction. The crystallographic orientations of the Ta₂C grains were homogenized by incorporating this relationship (Fig. 1c).

A clean-up procedure was carried out after this pseudosymmetry correction. In this procedure, the minimum grain size was defined as 6 pixels and the selected tolerance angle was 5°.

3. Experimental results

3.1. Top views of microstructures

Carburized samples are exhibiting a concentric structure on the holder side (Fig. 2), comprising of four zones with different microstructures (Fig. 3).

Fig. 3a) shows a top-view micrograph of Zone 1. It reveals a homogeneous surface with grain boundaries running across it. The surface is scattered with light patches. Fig. 3b) was taken in Zone 2. The surface is composed of oriented needles. Different regions of preferential orientations are visible in this micrograph. Each region appears to have its own preferential orientations, which are unrelated to one another.

The top-view micrograph of Zone 3 (Fig. 3c) shows a structure analogous to that of Zone 2 in other words, a surface comprising of oriented needles that form several regions of preferential orientations. However, the relief created by these needles seems to be much less distinct and much more homogeneous in Zone 3 than in Zone 2. Zone 4 (Fig. 3d) shows a completely different structure from the structures described earlier. It shows a surface of equiaxed grains with diameters in the micron range.

The diversity of the microstructures is the result of the variation in the associated carbon flow rate, which increases from the centre (Zone 1) to the edge of the surface (Zone 4). The next analyses were carried out in order to identify surface compounds and explain the orientations of certain microstructures.

3.2. Study of Zone 1

The XRD analysis of Zone 1 reveals the presence of two Ta peaks (JCPDS card 4-0788) that can be attributed to the underlying substrate. The analysis also shows that two compounds were detected: Ta₂C (JCPDS card 32-1280) and C3Ta4 (JCPDS card 03-065-3191). The compound C3Ta4 is described in the literature as having an NaCl structure of the [39] $Pm\bar{3}m$ space group with a lattice parameter of 4.42 Å. This compound is in fact TaC_{0.75} and therefore an under-stoichiometric TaC (TaC_{us}). TaC_{us} must not be confused with the compound Ta₄C₃, which has a rhombohedral structure and belongs to the $R\bar{3}m$ [21,40,41] space group.

Fig. 4a) shows a cross-sectional view of the structures that developed on the surface of Zone 1; a carbide layer is visible on this surface. This layer is polycrystalline and mainly composed of a single layer of grains. The carbide layer is $431 \pm 82 \text{ nm}$ thick and the grains are less than 500 nm wide. A precipitate located beneath the carbide layer is also visible. The parallel striations in this precipitate are indicative of significant twinning. Twinning in Ta₂C precipitates has already been reported by Dahmen [42]. According to this author, twinning is a mechanism for accommodating the stresses that are induced during the cooling stage.

The phase map (Fig. 4b) shows that the surface layer was indexed as Ta₂C. As only a negligible amount of TaC_{us} was indexed, this compound did not show up in the phase map.

The crystallographic orientation map (Fig. 4c) shows that the Ta₂C layer is composed of grains having different orientations. There appear to be three orientations. As the high indices of these planes do not give a clear indication of the orientations, the latter have been plotted in an inverse pole figure (Fig. 5).

Fig. 4c) also shows that the analyzed zone only contains a single Ta grain. The preferential growth direction of the Ta₂C layer and the small number of crystallographic orientations point to the existence of an epitaxial relationship between Ta and Ta₂C. Pole figures were plotted to explore the relationship between a Ta₂C grain (Fig. 6b) and a Ta grain (Fig. 6a), as well as that between the entire Ta₂C layer (Fig. 6d) and Ta (Fig. 6c). RD and TD stand for Rolling

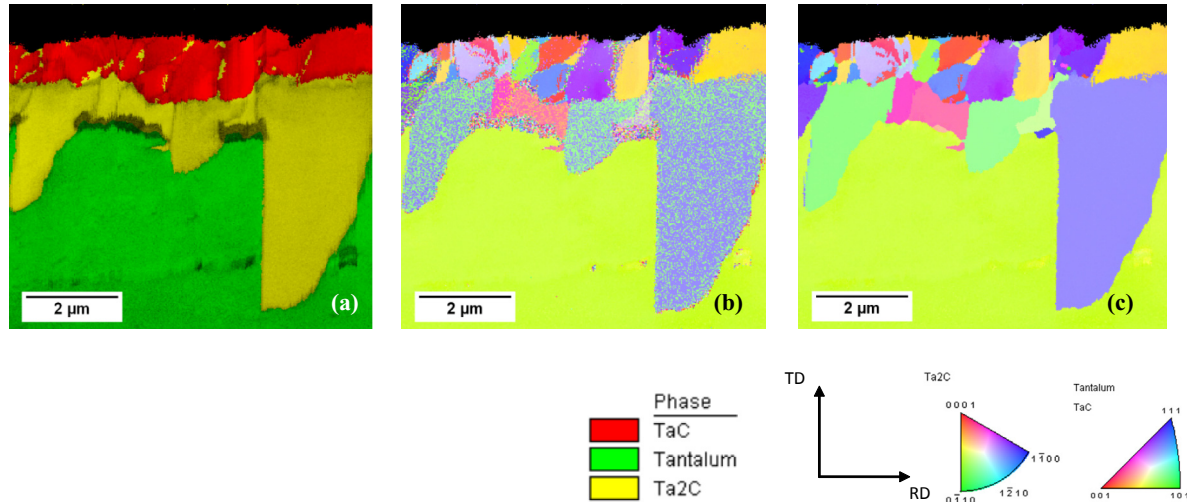


Fig. 1. Post-processing procedure: (a) analyzed phases + quality index; (b) crystallographic orientations, raw data; and (c) crystallographic orientations after pseudosymmetry corrections.

Direction and Transverse Direction respectively. These directions constitute the axes of the pole figures. Only the positive hemispheres have been represented.

The pole figure for the comparison between a Ta_2C grain (Fig. 6b) and a Ta grain (Fig. 6a) indicates an epitaxial relationship between two planes of the $\{10\bar{1}\}$ family in Ta and two planes of the $\{110\}$ family. Such a relationship was previously reported in the literature [42]. The $\{110\}$ pole figure obtained for the Ta_2C layer (Fig. 6d) shows that the Ta_2C grains have other orientations besides those shown in Fig. 6c). Some of these are planes belonging to the $\{10\bar{1}\}$ family in Ta. All the $\{10\bar{1}\}$ planes of the pole figure match with the $\{110\}$ planes in Ta_2C . This implies that the Ta_2C grains possess several crystallographic orientations, but the epitaxial relationship is maintained and may apply to other $\{10\bar{1}\}$ planes in Ta.

3.3. Study of Zone 2

The XRD analysis of Zone 2 indicates the presence of the same phases as those in Zone 1, namely Ta_2C and TaC_{US} . The Ta substrate was also detected. The results relating to Ta are comparable to those obtained for Zone 1. As the TaC_{US} peaks are all present in relation to JCPDS card 03-065-3191, the number of crystallographic orientations analyzed in the irradiated volume is higher than in Zone 1.

High-resolution micrography and EBSD analyses were carried out on cross-sections with the aim of explaining the preferential orientations of the surface microstructures.

The high-resolution SEM image (Fig. 7a) shows two phases at

the surface, connected by an intermediate phase. The surface phase is discontinuous and forms a series of small bumps. These bumps are composed of alternating parallel bands which are characteristic of twinning. The bands all lie perpendicular to the surface. The EBSD analysis of Zone 2 (Fig. 7b and c) shows that the surface phase is composed of TaC. The indexing software indexed the TaC_{US} detected by XRD as stoichiometric TaC because of the small difference in lattice parameters (TaC_{US} : 4.42 Å and TaC: 4.4545 Å). Most of the TaC grains have only two distinct orientations, which are shown in Fig. 8.

The second phase comprises highly elongated grains, the longest of which measure 8.5 µm from the surface (Fig. 7a). Each surface bump is associated with an elongated grain. The EBSD analysis (Fig. 7b and c) shows that these grains are composed of Ta_2C . The layer is composed of several columnar grains of the same orientation.

The intermediate phase is a transitional phase between the surface phase and the elongated phase. This phase is lamellar and lies perpendicular to the surface. The lamellae are oriented in the same direction as the twin boundaries. The intermediate phase was not indexed by the EBSD analysis. The EBSD analyses concur with the XRD analysis on the nature of the surface phases.

Pole figures were plotted to study the possible crystallographic relationship between the Ta substrate and the Ta_2C layer, as well as that between the Ta_2C and TaC layers. Fig. 9 a and b show the Ta $\{10\bar{1}\}$ and Ta_2C $\{110\}$ pole figures. Two $\{10\bar{1}\}$ planes in Ta correspond to two $\{110\}$ planes in Ta_2C (Fig. 9b). This configuration is identical to that obtained in Zone 1.

The crystallographic relationships between Ta_2C and TaC were also studied by means of pole figures (Fig. 9a and b). A comparison of the TaC $\{111\}$ and Ta_2C $\{0001\}$ pole figures (Fig. 9c and d) shows that:

- A plane from the $\{111\}$ family in TaC corresponds to $\{0001\}$ in Ta_2C .
- The TaC layer contains twins and the twin plane of TaC belongs to the $\{111\}$ family.

Indeed, the $\{111\}$ family possesses four equivalent planes. But the $\{111\}$ pole figure for TaC shows seven poles. Two distinct orientations may be observed in the pole figure. Therefore, there should be eight poles. However, two of the poles coincide because

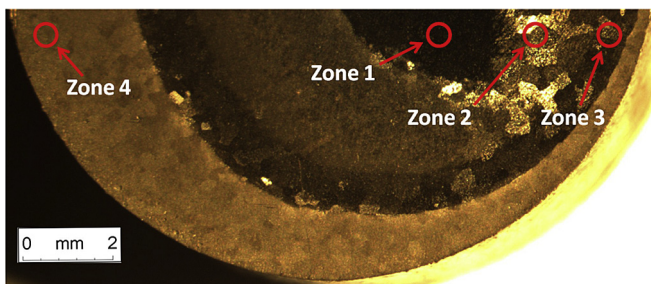


Fig. 2. Positions of the studied zones on the holder side.

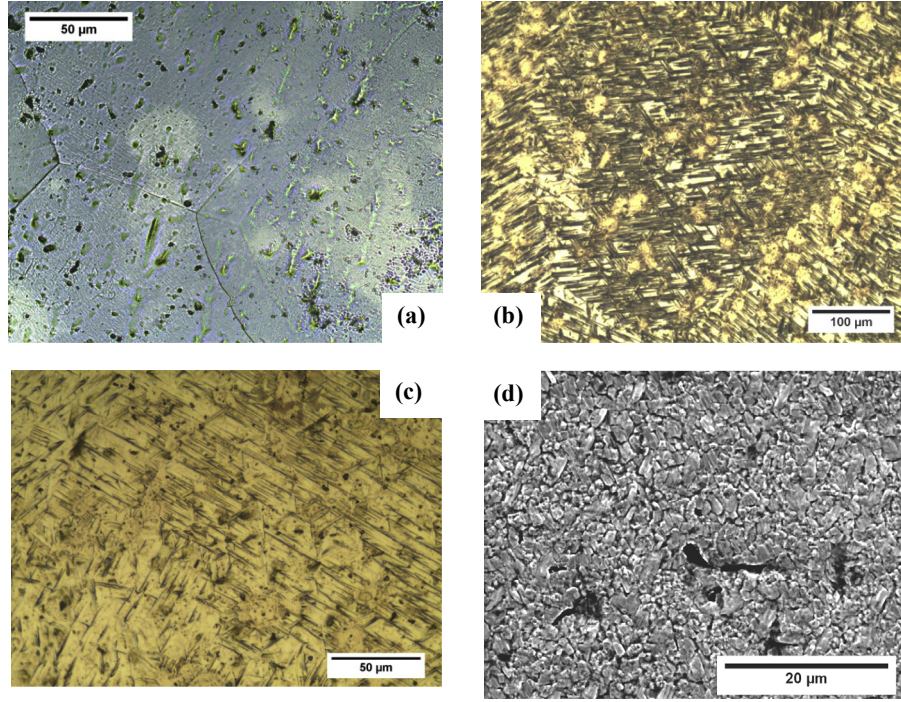


Fig. 3. Optical micrographs of zone 1 (a), zone 2 (b), zone 3 (c) and SEM micrograph of zone 4 (d).

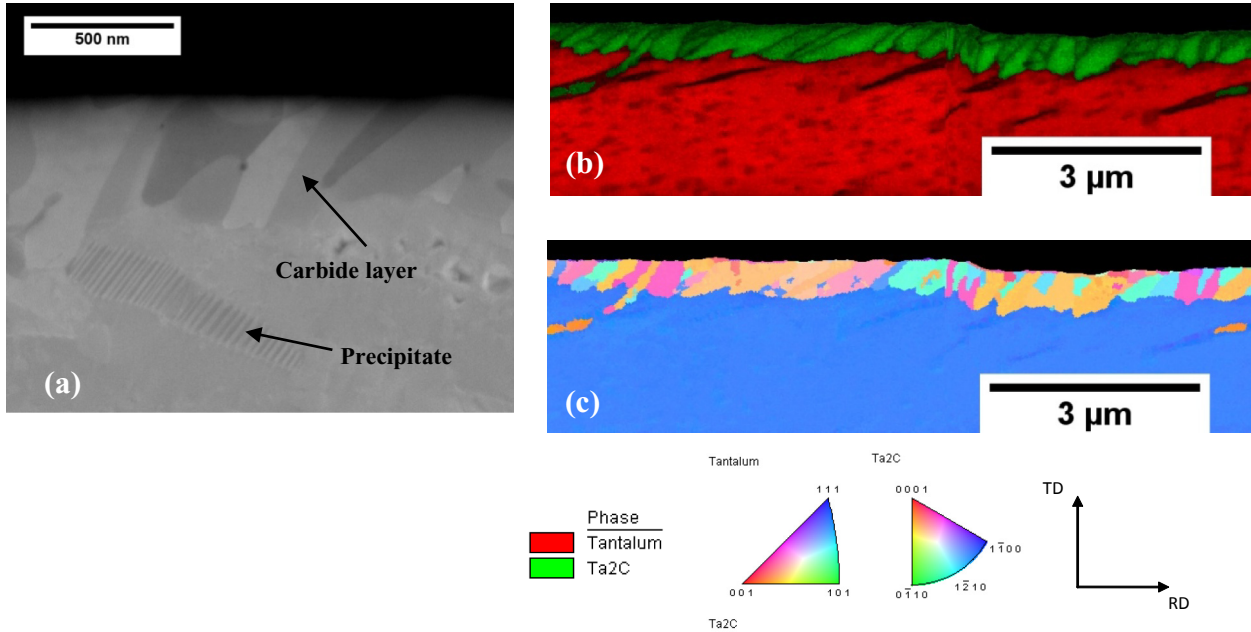


Fig. 4. (a) SEM micrograph of the cross-section of Zone 1 (backscattered electrons, 15 kV), (b) Phase + quality index map of the surface of Zone 1 and (c) IPF map of Zone 1.

the two orientations share a plane. This reduces the number of poles to seven.

Misorientation measurements were conducted between several twinned TaC grains. The measured angle was $60^\circ \pm 0.5^\circ$. This implies that the twinning in TaC consisted in a 60° rotation about the [111] axis.

All the above observations can be explained by assuming that the intermediate phase is composed of Ta₄C₃. According to Rowcliffe [43] and Morris et al. [23], Ta₄C₃ could be an intermediate

phase between TaC and Ta₂C. There is a crystallographic relationship between these three phases: $\{111\}_{\text{TaC}} // \{0001\}_{\text{Ta}_4\text{C}_3} // \{0001\}_{\text{Ta}_2\text{C}}$; $\langle 110 \rangle_{\text{TaC}} // \langle 10\bar{1}0 \rangle_{\text{Ta}_4\text{C}_3} // \langle 10\bar{1}0 \rangle_{\text{Ta}_2\text{C}}$.

This relationship accounts for why the TaC twin boundaries are aligned with the Ta₄C₃ laths (Fig. 7a).

Thus, it appears that the strong orientation of the TaC needles results from the relationships between TaC and Ta₂C. There is also a relationship between the growth of Ta and Ta₂C. It is therefore likely that the orientation of the needles is directly linked to the

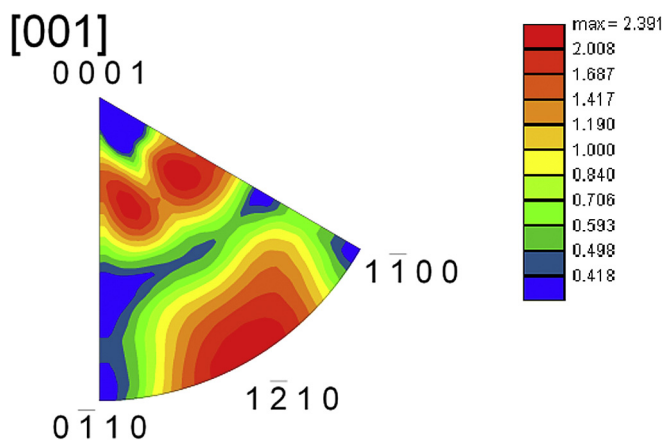


Fig. 5. [001] IPF showing the preferential orientations of the Ta₂C layer in Zone 1.

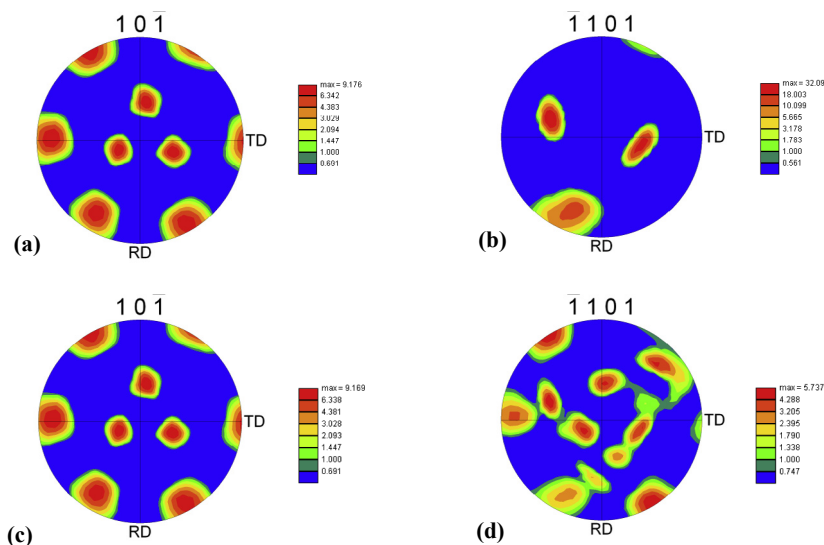


Fig. 6. Pole figures obtained for Zone 1: $\{10\bar{1}\}/\{10\bar{1}\}$ pole figure for Ta, $\{\bar{1}101\}/\{\bar{1}71010\}$ pole figure for Ta₂C. Comparison between (a) a Ta grain and (b) a Ta₂C grain. Comparison between (c) a Ta grain and (d) the entire Ta₂C layer.

orientation of the Ta grains that make up the substrate of the carbide layers. This accounts for the texture followed by the needle growth (Fig. 3b).

An examination of the pole figures also reveals twinning in the TaC layer by a 60° rotation about the $\langle 111 \rangle$ axis. This relationship can explain the star-shaped structures formed by the TaC needles, as seen in Zone 2 (Fig. 3b).

3.4. Study of Zone 3

The XRD analysis of Zone 3 indicates the presence of the same phases as those in Zones 1 and 2, namely Ta₂C and TaC_{us}. The high-resolution (cross-sectional) micrograph of Zone 3 (Fig. 10a) shows a two-layer structure composed of two distinct phases connected by an intermediate lamellar phase. The surface phase is composed of columnar grains. The EBSD analysis of Zone 3 (Fig. 10b) shows that the surface phase is composed of TaC. This layer is thicker than that measured in Zone 2 (thickness of less than 2 μm). The TaC layer (Fig. 10b) is composed of columnar grains with two distinct orientations. These two orientations seem to be identical to those in Zone 2. The second layer of grains underlying the TaC layer is composed of columnar grains. The grain boundaries of the TaC layer

extend into the underlying layer of grains. The EBSD analysis shows that this layer is composed of Ta₂C (Fig. 10b). The orientation of the Ta substrate seems to be identical to that in Zone 2. It is possible that the EBSD analyses of Zones 2 and 3 were carried out in the same Ta grain. This would account for the similarities in the orientations within the TaC layer of these two zones.

The phase and quality index map (Fig. 10b) shows that the Ta₂C layer is composed of grains having a columnar structure. The quality index map shows distinct grain boundaries in the Ta₂C layer. However, the crystallographic orientation map (Fig. 10c) shows that the Ta₂C layer is only composed of two grains that display pseudosymmetry relationship. The Ta₂C grain boundaries shown in Fig. 10b are not visible in the crystallographic orientation map. This is due to the pseudosymmetry correction procedure applied to Ta₂C. The raw data indicate the existence of a low level of misorientation between the various Ta₂C grains. The misorientation was measured between 0 and 8°, which suffices to account for the

presence of grain boundaries in the quality index map. This misorientation was eliminated by the pseudosymmetry correction.

Pole figures for Zone 3 show the same results than zone 2. The epitaxial relationship between Ta $\{10\bar{1}\}$ and Ta₂C $\{\bar{1}101\}$ and between TaC $\{111\}$ and Ta₂C $\{0001\}$ was verified.

3.5. Study of Zone 4

A XRD analysis of Zone 4 was obtained for the area containing the highest amount of carbon. The analysis shows the appearance of a 'new' phase, stoichiometric TaC (JCPDS card 35-0801). This 'new' phase can be attributed to an enrichment of TaC_{us} with carbon, causing an increase in the lattice parameter [6] and a shift in the peaks towards smaller angles.

Fig. 11a) shows a cross-sectional view of the surface structure in Zone 4. The layer that is visible in this micrograph comprises two layers of grains. The EBSD analysis (Fig. 11b) shows that these correspond to a TaC surface layer and an underlying Ta₂C layer. The nature of the phases is identical to those identified in Zones 2 and 3. However, the EBSD analysis of Zone 4 reveals a visible change in the microstructure of the surface layers with respect to the layers in Zones 2 and 3. While the phase and quality index map only shows a

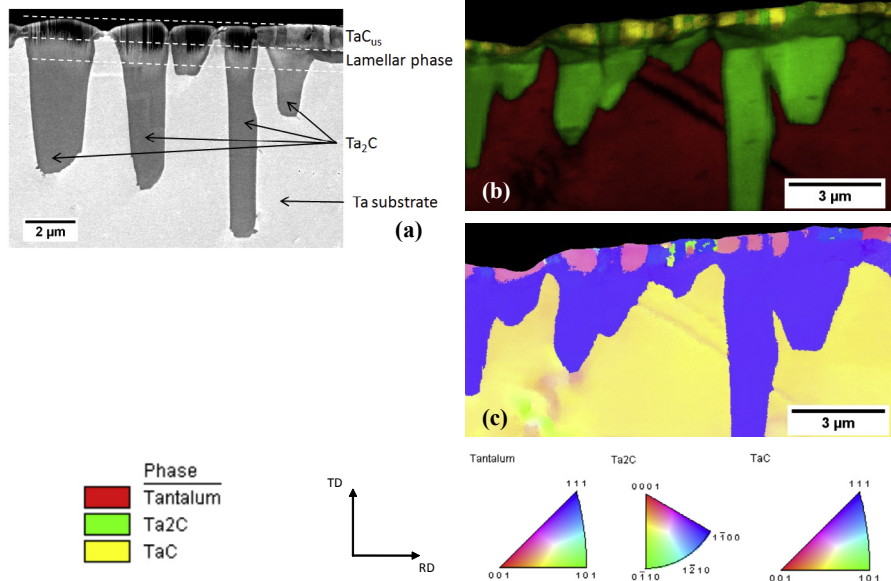


Fig. 7. (a) High-resolution micrograph of the cross-section of Zone 2 (backscattered electrons, 15 kV). (b) Phase + quality index map of the surface of Zone 2 and (c) crystallographic orientation map of Zone 2.

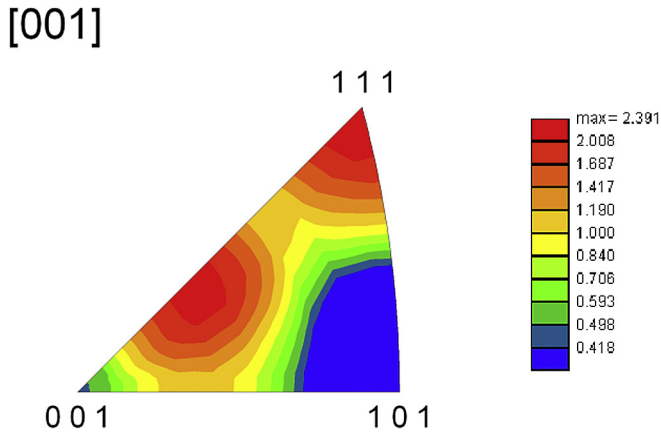


Fig. 8. [001] IPF showing the preferential orientations of the TaC layer in Zone 2.

thickening of the TaC layer at the expense of the Ta₂C layer (Fig. 11b), the crystallographic orientation map (Fig. 11c) indicates a real change in the orientations of the two layers. The TaC and Ta₂C layers possess a much wider range of crystallographic orientations.

However, it is important to note that Fig. 11b) shows the columnar structure of the two carbide layers and a continuity between the TaC and Ta₂C grain boundaries. Also, there are only a few grain boundaries between the TaC and Ta₂C layers. This may be attributed to the presence of the intermediate phase that is visible in Fig. 11a). This intermediate phase was not indexed.

The Ta {101} pole figure (Fig. 12a) and the Ta₂C {1101} pole figure (Fig. 12b) confirm the diversification of the crystallographic orientations of Ta₂C. There is much more mismatch between the Ta and Ta₂C poles.

The TaC {111} pole figure (Fig. 12c) and the Ta₂C {0001} pole figure show that the TaC and Ta₂C poles still match, which proves that there are still certain crystallographic relationships between these two phases.

In order to verify this, other pole figures were plotted for areas including a TaC grain (Fig. 13a) and an adjacent Ta₂C grain (Fig. 13b).

The pole figures show that both TaC grains experienced slight twinning. They also prove the existence of a relationship between the {0001} plane in Ta₂C and one of the {111} planes in TaC.

It appears that the crystallographic relationships between Ta and Ta₂C in Zone 4 are not as strong as those in the other zones. But the relationships between TaC and Ta₂C still exist.

The diversity of the crystallographic orientations of the TaC layer can account for the homogeneous and equiaxed microstructure that is visible in the surface micrograph (Fig. 3d).

4. Discussion

4.1. Cross-sectional views of microstructures

The carbon flux provided by the containment technique is not well defined. However, the flux is clearly reduced in the holder side compared to the chamber side. Optical micrographs were carried out on both sides (Fig. 14). They show that the Ta₂C layer is thinner in holder side than in chamber side. However, carbide layers seem to be thicker in holder side (zone 5, Fig. 14b) than in chamber side (Fig. 14c). But the TaC layer is thicker in chamber side. Therefore, carbide layers in holder side contain less carbon despite they seem thicker.

The increase in carbon flow from the centre (Zone 1) to the edge (Zone 4) of the sample resulted in the progressive carbon enrichment of the surface. The small amount of TaC_{US} detected in Zone 1 increased in Zones 2 and 3, and even more so in Zone 4. It also became enriched with carbon and eventually turned into stoichiometric TaC. This variation in the nature of the surface tantalum carbide can be explained by the degree to which the tantalum became saturated with carbon. The thinnest carbide layer (431 ± 82 nm) was found in Zone 1. This layer was composed of Ta₂C, the carbide having the lowest carbon content in the Ta–C system [16]. This indicates that the carbon inflow was barely greater than the carbon diffusion rate close to the tantalum surface. The carbon flow rate was just sufficient to saturate the tantalum surface with carbon and thus create the Ta₂C layer. The increase in the carbon flow in Zone 2 accelerated the growth of the Ta₂C grains, leading to the appearance of highly elongated grains in the Ta

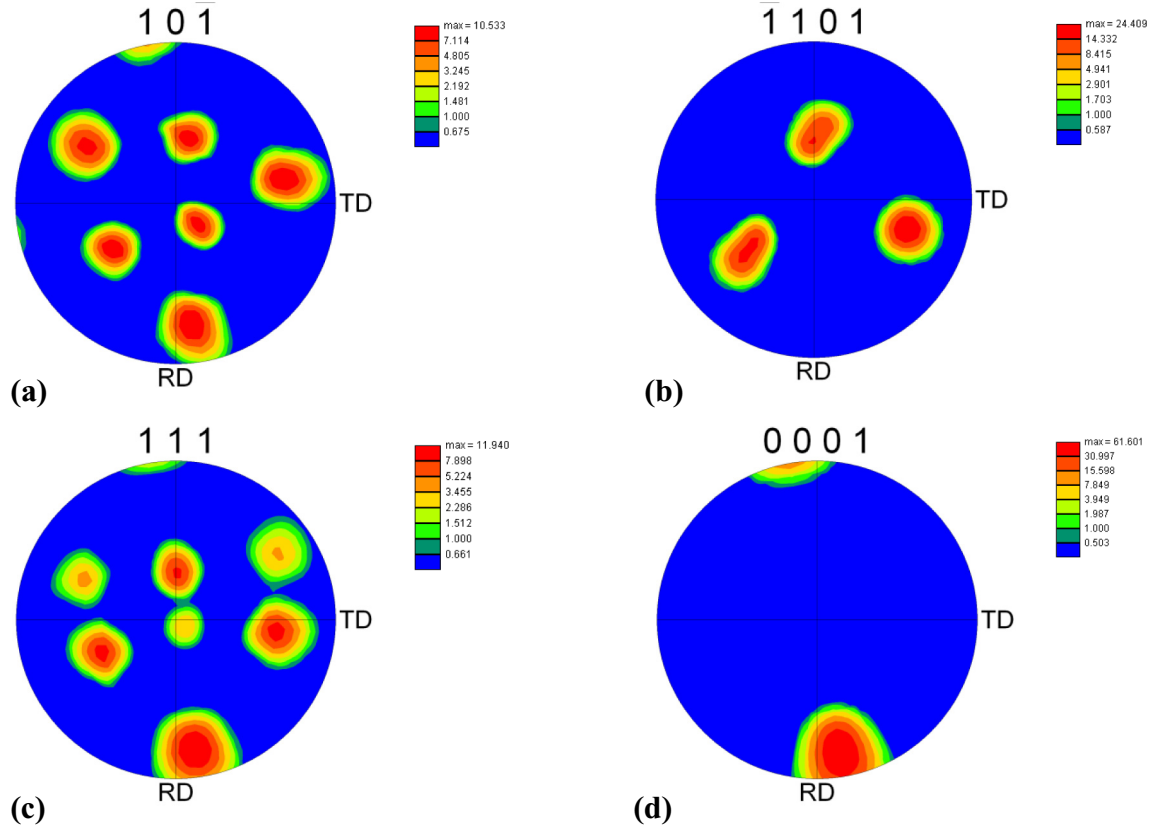


Fig. 9. Pole figures obtained for Zone 2. (a) $\{10\bar{1}\}/\langle 10\bar{1} \rangle$ pole figure for Ta, (b) $\{\bar{1}101\}/\langle \bar{1}101 \rangle$ pole figure for Ta₂C, (c) $\{111\}/\langle 111 \rangle$ pole figure for TaC and (d) $\{0001\}/\langle 0001 \rangle$ pole figure for Ta₂C.

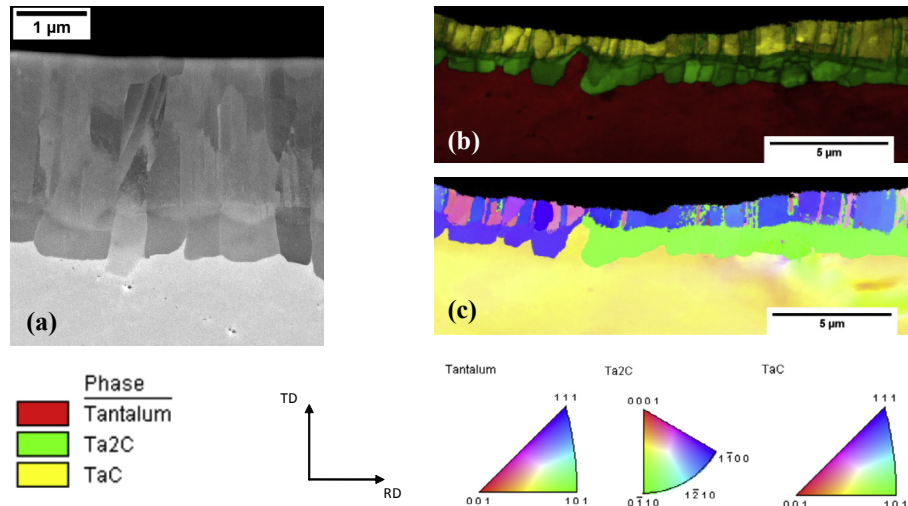


Fig. 10. (a) High-resolution micrograph of the cross-section of Zone 3 (backscattered electrons, 15 kV). (b) Phase + quality index map of the surface of Zone 3 and (c) crystallographic orientation map of Zone 3.

substrate. The carbon flow rate was also sufficient to saturate the Ta₂C layer with carbon, leading to the appearance of TaC on the surface. To our knowledge, the highly irregular growth of Ta₂C grains has not been accounted for in the literature. It is possible that this exaggerated growth is due to a specific crystallographic orientation of Ta₂C. An examination of the $\{0001\}/\langle 0001 \rangle$ pole figure for Ta₂C (Fig. 9d) shows that the $\{0001\}$ planes in the Ta₂C grains are almost perpendicular to the surface, along the axis of

grain growth. Meanwhile, Dahmen [42] has affirmed on his studies involving the precipitation of Ta₂C in a Ta matrix that the favoured direction of growth of Ta₂C precipitates is parallel to the $\{0001\}$ planes. Thus, there is likely to be a connection between the crystallographic orientation of the Ta₂C grains and their exaggerated growth (in comparison with the other zones). This hypothesis still needs to be confirmed. Compared with Zones 1 and 2, the increased carbon flow in Zone 3 led to an earlier appearance of the TaC layer.

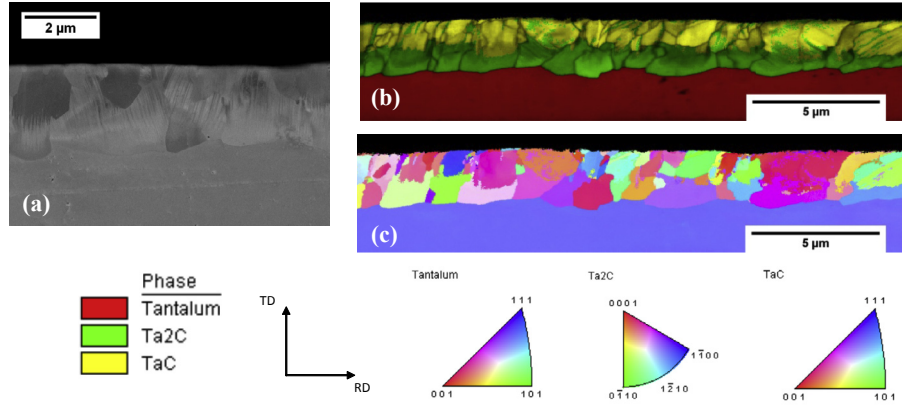


Fig. 11. (a) High-resolution micrograph of the cross-section of Zone 4 (backscattered electrons, 15 kV), (b) Phase + quality index map of the surface of Zone 4 and (c) crystallographic orientation map of Zone 4.

At 1600 °C, the diffusion coefficient of carbon was lower in the TaC layer than in the Ta₂C layer [7,26]. The TaC layer acted therefore as a diffusion barrier against carbon, thereby hindering the growth of the Ta₂C layer.

The carbon flow rate seems to govern the growth mode of the tantalum carbide layers. The columnar and oriented growth of the TaC and Ta₂C grains in Zones 2 and 3 is much less visible in Zone 4, which experienced the highest carbon flow. The crystallographic orientation map of Zone 4 (Fig. 11c) shows a wide range of crystallographic orientations in the Ta₂C layer, which is confirmed by the pole figures for Ta and Ta₂C. The crystallographic relationships between TaC and Ta₂C are nonetheless maintained. There appears to be a carbon flow 'limit', beyond which the columnar growth of carbide grains is altered or even eliminated.

4.2. Top views of microstructures

The top views of the surface microstructures also vary with the

carbon flow. The top-view surface microstructure of Zone 2 (Fig. 3b) is composed of TaC needles and islets. The nature of the surface phase was determined through EBSD analyses. The influence of the carbon supply rate on the morphology of the surface microstructure has been addressed in the literature. Xiong et al. published a study on the epitaxial growth of TaC on graphite by CVD deposition [44]. They show that the rate of TaC deposition greatly influences the morphology of the carbide layer, which varies between a columnar and pyramidal structure at low deposition rates and a cauliflower structure at higher deposition rates. The present study points a carburizing stage which comprised two steps: carbon enrichment of the atmosphere (with a hydrogen + solid carbon combination), followed by dilution of the carbonaceous atmosphere by means of a primary vacuum. The supply of carbon to the tantalum therefore decreased when the vacuum was produced, during the carburizing stage. The presence of two different structures (islets and needles) composed of TaC within the same zone can therefore be attributed to the variation in the carbon flow

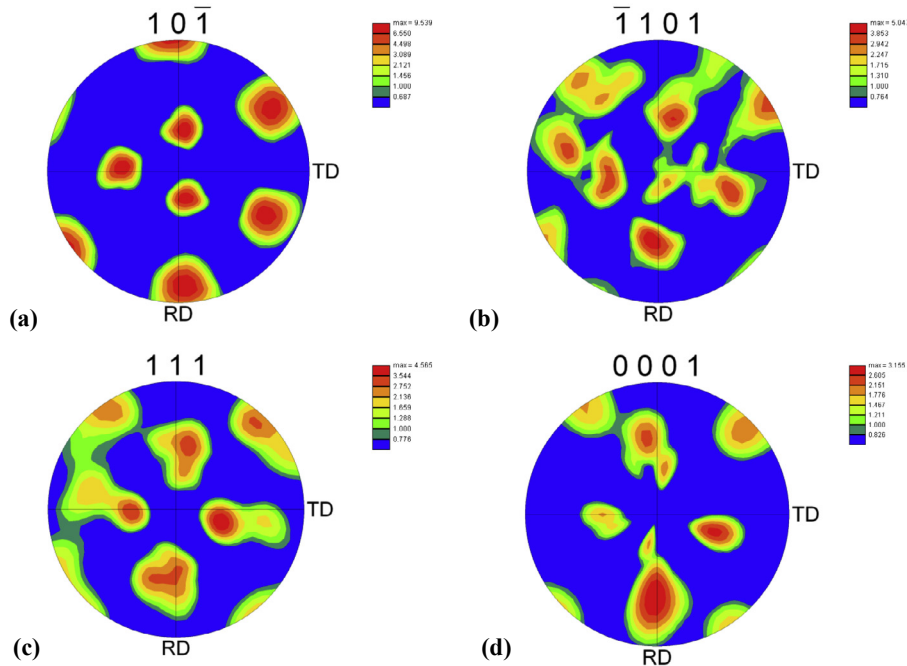


Fig. 12. Pole figures obtained for Zone 4. (a) $\{10\bar{1}\}/\{10\bar{1}\}$ pole figure for Ta, (b) $\{\bar{1}101\}/\{\bar{1}17010\}$ pole figure for Ta₂C, (c) $\{111\}/\{111\}$ pole figure for TaC and (d) $\{0001\}/\{0001\}$ pole figure for Ta₂C.

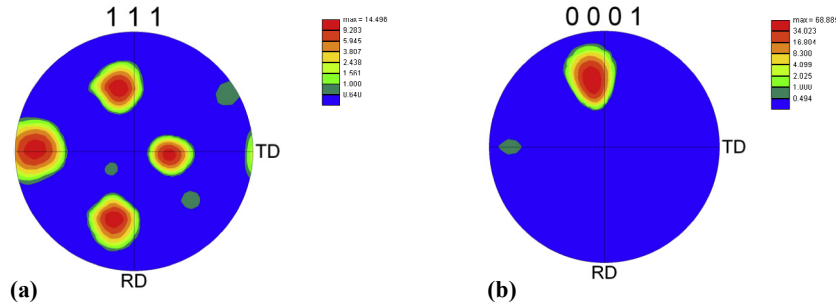


Fig. 13. Pole figures obtained for adjacent TaC and Ta₂C grains, (a) {111} pole figure for the TaC grain, (b) {0001} pole figure for the Ta₂C grain.

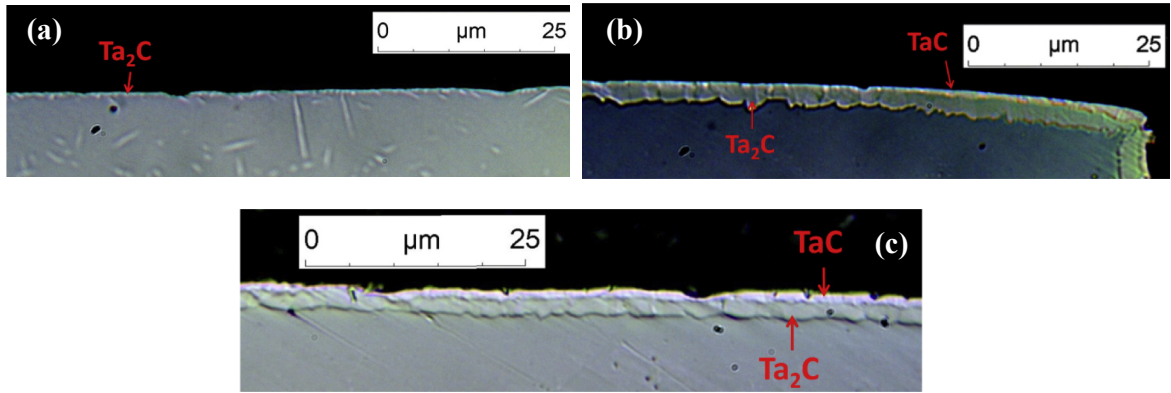


Fig. 14. (a) Optical micrograph of the holder side in zone 1, (b) in zone 5, (c) chamber side.

during the carburizing treatment.

The idea of a flow limit mentioned in the preceding paragraph becomes much clearer when the top views of Zones 2 and 3 (Fig. 3b and c) and Zone 4 (Fig. 3d) are compared. The strong texture of Zones 2 and 3 is replaced by an equiaxed microstructure in Zone 4.

5. Conclusions

The carburizing of a tantalum sample under different carbon flow rates has highlighted the influence of the carbon flow rate on the surface microstructures.

A global analysis of the EBSD results has led to several conclusions. The first conclusion is that for a certain carbon flow rate, various tantalum carbide layers may grow epitaxially on the Ta substrate. Tantalum is related to Ta₂C by the following crystallographic relationship: Ta {10 $\bar{1}$ } // Ta₂C { $\bar{1}$ 101}; <10 $\bar{1}$ >_{Ta} // < $\bar{1}$ 717010>_{Ta₂C}. The TaC layer is connected to the Ta₂C layer by a third intermediate Ta₄C₃ phase. The relationship between TaC and Ta₂C [1,23] has been verified through the use of pole figures. This relationship is as follows: {111}_{TaC} // {0001}_{Ta₄C₃} // {0001}_{Ta₂C}; <110>_{TaC} // <10 $\bar{1}$ 0>_{Ta₄C₃} // <10 $\bar{1}$ 0>_{Ta₂C}. The very strong texture of the microstructure in Zones 2 and 3 is the result of these crystallographic relationships.

The second conclusion is that beyond a certain carbon flow limit, the columnar and textured growth of the carbide layers is halted. Below this carbon flow limit, the carbide layer growth is columnar and highly textured because of the strong crystallographic relationships between the various phases. Exceeding the carbon flow limit has a strong impact on the morphology of the TaC layer. The grains are much more equiaxed and the development of oriented microstructures at the surface is interrupted. The crystallographic relationships between TaC and Ta₂C are nonetheless

maintained, but those between Ta and Ta₂C are not as clear-cut. The actual value of the carbon flow limit has not been obtained. Determining this value would require flow calculations that are difficult to perform analytically.

It is interesting to note that epitaxial surface structures on substrates are generally obtained through CVD [44], PVD [45] or PLD [46]. It is therefore possible to obtain epitaxial tantalum carbides on substrates by low-pressure carburizing, with a device that limits the carbon flow towards the tantalum.

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References

- [1] D.J. Rowcliffe, W.J. Warren, Structure and properties of tantalum carbide crystals, *J. Mater. Sci.* 5 (1970) 345–350.
- [2] A.S. Audion, P. Piantone, Panorama 2011 du marché du tantale, BRGM, rapport interne RP-61343-FR, 2012.
- [3] C.F. Zalabak, The Melting Points of Tantalum Carbide and of Tungsten, Technical report No. NASA-TN-D-761, 1961.
- [4] G. Santoro, H.B. Probst, An explanation of microstructures in the tantalum-carbon system, *Adv. X-Ray Anal.* 7 (1963) 126–135.
- [5] E. Rudy, D.P. Harmon, Technical report No. AFML-TR-65-2, part I, Ternary Phase Equilibria in Transition Metal-boron-carbon-silicon Systems, vol. V, 1965.
- [6] E.K. Storms, *The Refractory Carbides*, Academic Press, New York and London, 1967, pp. 82–93.
- [7] W.F. Brizes, Diffusion of carbon in the carbides of tantalum, *J. Nucl. Mater.* 26 (1968) 227–231.

- [8] E. Fromm, U. Roy, The high-temperature solid solubility limit of carbon in tantalum, *J. Less Common Met.* 8 (1965) 73–75.
- [9] B.H. Eckstein, R.F. Forman, Preparation and some properties of tantalum carbide, *J. Appl. Phys.* 33 (1962) 82–87.
- [10] G. Hörz, K. Lindenmaier, The kinetics and mechanisms of the absorption of carbon by niobium and tantalum in a methane or acetylene stream, *J. Less Common Met.* 35 (1974) 85–95.
- [11] G. Hörz, K. Lindenmaier, Carburization of refractory metals in hydrocarbons, in: 4. annual Spring Meeting of the Metallurgical Society of AIME, Boston, Massachusetts, USA, 8 May 1972, 1973, pp. 168–181.
- [12] G. Santoro, Variation of some properties of tantalum carbide, *Trans. Metall. Soc. AIME* 227 (1963) 1361–1368.
- [13] A. Tardif, J.M. Michel, J. Wach, Cinétique de décomposition du pentane sous $10^{-5}/10^{-7}$ torr à la surface du tantale et des carbures Ta_2C et TaC , entre 1300 et 2300K, *Surf. Sci.* 26 (1971) 255–268.
- [14] A. Raveh, A. Danon, J. Hayon, A. Rubinshtein, R. Shneck, J.E. Klemberg-Sapieha, L. Martinu, Characterisation of carburized tantalum layers prepared in inductive RF plasma, *Thin Solid Films*, 392 (2001) 56–64.
- [15] A. Rubinshtein, R. Shneck, A. Danon, J. Hayon, S. Nathan, A. Raveh, Surface treatment of tantalum to improve its corrosion resistance, *Mater. Sci. Eng. A* 302 (2001) 128–134.
- [16] F.H. Ellinger, The tantalum - carbon system, *Trans. Am. Soc. Met.* 31 (1943) 89–104.
- [17] R. Lesser, G. Brauer, Karbidphasen des Tantals, *Z. für Met.* 49 (1958) 622–626.
- [18] E. Rudy, El Rudy, F. Benesovsky, Untersuchungen in system tantal-wolfram-kohlenstoff, *Monatsh. für Chem. Verwandte Teile Anderer Wiss.* 93 (1962) 1176–1195.
- [19] R.V. Sara, C.E. Lowell, R.T. Dolloff, Research Study of Determine the Phase Equilibrium Relations of Selected Metal Carbides at High Temperatures, Technical report No. WADDTR60-143, part IV, 1963.
- [20] A.L. Bowman, The crystal structures of V_2C and Ta_2C , *Acta Crystallogr.* 19 (1965) 6–9.
- [21] A.I. Gusev, A.S. Kurlov, V.N. Lipatnikov, Atomic and vacancy ordering in carbide Zeta Ta_4C_{3-x} (0.28×0.4) and phase equilibria in the Ta-C system, *J. Solid State Chem.* 180 (2007) 3234–3246.
- [22] M. Rocher, P. Goeuriot, J. Dhers, Modelling of the growth of carbide layers in tantalum, *Key Eng. Mater.* 206–213 (2002) 527–530.
- [23] R.A. Morris, B. Wang, L.E. Matson, G.B. Thompson, Microstructural formations and phase transformation pathways in hot isostatically pressed tantalum carbides, *Acta Materialia*, 60 (2012) 139–148.
- [24] V.N. Lipatnikov, A.A. Rempel, Formation of the incommensurate ordered phase in TaC_x carbide, *J. Exp. Theor. Phys.* 81 (2005) 326–330.
- [25] A.I. Gusev, Sequence of phase transformations in the formation of superstructures of the M_6C_5 type in nonstoichiometric carbides, *J. Exp. Theor. Phys.* 109 (2009) 417–433.
- [26] R. Resnick, R. Steinitz, L. Seigle, Determination of diffusivity of carbon in tantalum and columbium carbides by layer-growth measurement, *Trans. Metall. Soc. AIME* 233 (1965) 1915–1918.
- [27] I. Zaplatynsky, Observations on zeta phase in the system Ta-C, *J. Am. Ceram. Soc.* 49 (1966) 109–110.
- [28] E. Fromm, E. Gebhardt, U. Roy, Diffusion des Kohlenstoffs in den Karbidphasen des Tantals, *Z. für Met.* 57 (1966) 808–811.
- [29] N.H. Krikorian, T.C. Wallace, R. Krohn, M.G. Bowman, Formation of Carbide Surfaces on Tantalum and Tantalum-tungsten Alloys, Los Alamos National Laboratory, 1968. LA-AC-8803.
- [30] H. Wiesenberger, W. Lengauer, P. Ettmayer, Reactive diffusion and phase equilibria in the V-C, Nb-C, Ta-C and Ta-N systems, *Acta Materialia* 46 (1998) 651–666.
- [31] W. Lengauer, Multiphase reaction diffusion in transition metal-carbon and transition metal-nitrogen systems, *J. Alloys Compd.* 229 (1995) 80–92.
- [32] M. Rocher, Carburation d'un métal réfractaire et de ses alliages, Mécanismes et Modélisation, Ecole Nationale Supérieure des Mines de Saint Etienne, France, 2002. Thèse N°274TD.
- [33] M. Balooch, D.R. Olander, Reactions of modulated molecular beams with pyrolytic graphite. III. Hydrogen, *J. Chem. Phys.* 63 (1975) 4772–4786.
- [34] E. Hoinkis, The chemisorption of hydrogen on porous graphites at low pressure and at elevated temperature, *J. Nucl. Mater.* 182 (1991) 93–106.
- [35] R.A. Krakowski, D.R. Olander, Survey of the Literature on the Carbon-hydrogen System, Technical report LBNL UCRL-19149, 1970.
- [36] D. Cotton, Etude de formation d'hémicarbure de tantale (Ta_2C) par l'intermédiaire d'un procédé de cémentation sous pression réduite, PhD Thesis tel-01213990, Université de Bourgogne, 2015.
- [37] M.M. Nowell, S.I. Wright, Orientation effects on indexing of electron backscatter diffraction patterns, *Ultramicroscopy* 103 (2005) 41–58.
- [38] P. Vonlanthen, B. Grobety, CSL grain boundary distribution in alumina and zirconia ceramics, *Ceram. Int.* 34 (2008) 1459–1472.
- [39] V.G. Zubkov, L.B. Dubrovskaya, P.V. Gel'd, V.A. Tskhai, Y.A. Dorofeev, Neutron-diffraction Study of the Ordering of Non Stoechiometric Cubic Carbides of Fifth-group Transition Metals, vol. 184, *Doklady Akademii Nauk SSSR*, 1968, pp. 874–876.
- [40] V.N. Lipatnikov, A.I. Gusev, Atomic vacancy ordering in the carbide phase Ta_4C_3 , *Phys. Solid State* 48 (2006) 1634–1645.
- [41] K. Yvon, E. Parthé, On the crystal chemistry of the close-packed transition-metal carbides. I. The crystal structure of the zeta-V, Nb and Ta carbides, *Acta Crystallogr. B* 26 (1970) 149–153.
- [42] U. Dahmen, Microstructures and Phase Transformations in Interstitial Alloys of Tantalum, Lawrence Berkeley National Laboratory, 1979. LBNL Paper LBL-8661.
- [43] D.J. Rowcliffe, Structure of non-stoichiometric TaC, *Mater. Sci. And Eng. A* 18 (1975) 231–238.
- [44] X. Xiong, Z. Chen, B. Huang, G. Li, F. Zheng, P. Xiao, H. Zhang, Surface morphology and preferential orientation growth of TaC crystals formed by chemical vapor deposition, *Thin Solid Films*, 517 (2009) 3235–3239.
- [45] T. Naiki, M. Ninomiya, M. Ihara, Epitaxial growth of tantalum carbide, *Jpn. J. Appl. Phys.* 11 (1972) 1106–1112.
- [46] R.D. Vispute, S. Hullavarad, A. Luykx, D. Young, S. Dhar, T. Venkatesan, Epitaxy and recrystallization kinetics of TaC thin films on SiC for high temperature processing of semiconductor devices, *Appl. Phys. Lett.* 90 (2007), 24917-1–24917-3.