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Damping Behavior in a Wide Temperature Range of FeMn-Like High Entropy Shape Memory Alloys

L. Peltier¹ · P. Lohmuller² · F. Meraghni¹ · E. Patoor³ · P. Laheurte² · S. Berveiller¹

Abstract In this work, a new class of iron-based high entropy shape memory alloys (HE-SMAs) have been designed, characterized, and optimized. These FeCu-NiMnV alloys (FeMn-like) with damping properties at low and high temperatures are developed; a methodology is proposed to demonstrate how to preserve this effect from the memory loss observed on conventional damping alloys. The developed alloys are analyzed using X-ray diffraction, scanning electron microscopy and differential scanning calorimetry. Their damping capacity is investigated using a drop weight test device instrumented with a digital image correlation system for the displacement measurement. It is compared with one of FeMnV, NiTi, CuAlNi, and 1050A alloys at different temperatures. The results show that the damping capacity is interesting over a wide range of operating temperatures. It has been established that the cocktail effect obtained by mixing Cu, Fe, Ni, Mn, and V elements allows for optimizing the damping capacity of the HE-SMAs. In addition, the sluggish diffusion may allow

these HE-SMAs to prevent the premature aging that leads to a degradation of the damping behavior notably at high temperature. The variation of the composition of the $\text{Cu}_x\text{Fe}_y\text{Ni}_z\text{Mn}_{20}\text{V}_{11}$ alloys enables the adjustment of the alloying element content by favoring the appearance of the non-thermally activated martensite keeping hence a stable damping behavior from -40 to 200 °C.

Keywords High entropy alloys · Shape memory alloys · Damping behavior · Martensitic transformation · Cold forming

Introduction

Research on high entropy alloys (HEAs) is extensively conducted during the last decade due to the high mechanical potential of this new class of alloys [1]. HEAs are generally defined as alloys containing at least five elements, with equimolar or near equimolar compositions [2]. The advantages of high entropy are high hardness level, high strength when used at elevated temperatures and a sensitive increase of corrosion and wear resistance [3]. Moreover, HEAs generally form a simple solid solution (FCC or BCC) at the detriment of the formation of intermetallic phases [4]. The element composition of the alloy is an important factor affecting its behavior and performance [5].

Fe-based and Cu-based shape memory alloys are very good candidates for civil applications requiring damping capabilities [6, 7]. Several families of SMAs are known for their damping effect: MnCu-type alloys [8–10], FeMn-type alloys [11–16], NiMn-type alloys [17, 18], NiTi-type alloys [19], CuAl-type alloys [20], and TiNb-type alloys [21]. Some of these families have known deficiencies such as

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✉ L. Peltier
laurent.peltier@ensam.eu

¹ LEM3, Arts et Métiers Institute of Technology, Université de Lorraine, CNRS, HESAM Université, 4, rue Augustin Fresnel, 57070 Metz, France

² LEM3, Université de Lorraine, Arts et Métiers Institute of Technology, CNRS, 7, rue Félix Savart, 57070 Metz, France

³ GT-UMI 2958 Georgia Tech-Lorraine, CNRS, 2 rue Marconi, 57070 Metz, France

loss of memory effect (or amnesia) for MnCu and CuAl [22–24] type alloys or too much temperature dependency for NiTi and CuAl-type alloys [25–28]. Jensen et al. [29] even show a significant loss of the damping effect of the binary MnCu alloy during long and unstressed storage. These features are unsuitable for industrial applications and their related reliability. The quaternary alloys M2052 (MnNiCuFe) [30–32] and SONOSTRON (MnCuFeAl) [33] known for their damping effect already appear as the mixing of several families of SMAs but are also subject to the degradation of the damping capacity observed during their storage [34]. Some recent studies already show the possibility of producing quinary alloys with a damping behavior whose entropy is high but does not correspond to the high entropy criterion [14, 35–38]. Other studies are investigating the behavior of HEAs with damping characters doped with oxygen and nitrogen [39].

This study aims at producing new high entropy shape memory FeMn-like alloys (HE-SMA) derived from the ternary Fe₆₇Mn₂₂Cr₁₁ alloy developed by Okada et al. [40]. The damping capabilities shall be valid over a wide temperature range (– 40 to 200 °C) and the microstructure shall allow a high ductility and improved cold formability.

The novelty of this paper resides in the methodology enabling to sum up the effects related to the high entropy to build a new type of FeMn-like damping alloys. The pseudoelastic effect, which is very low subject to temperature is generated by the as-cast structure of the alloy and the optimized internal compressive stresses due to the crystal lattice distortion. On the other hand, to the best of the authors' knowledge, there has been no comparative work on the simultaneous effects of high entropy on the manufacturing formability and the aging of shape memory damping alloys.

This paper is organized as follows: in section “[Choosing the Chemical Composition of the Studied Alloys](#)”, the choice of the five elements constituting the studied HEAs and the weighting of these five elements is presented and detailed using the equivalence parameters and the criteria related to the metallurgy of the HEAs allowing to transform a low entropy SMA into a high entropy SMA (HE-SMA). Section “[Experimental Procedure](#)” is devoted to the melting, heat treatment processing, and the test bench to evaluate damping properties. In section “[Experimental Results](#)”, the results of the different tests are presented, depending on the test temperature and the metallurgical state of the alloys. Test results of the four HE-SMAs produced for this work are then compared with those of a ternary alloy from which these HE-SMAs are derived, as well as to those of a NiTi alloy and a ternary CuAlNi alloy.

Choosing the Chemical Composition of the Studied Alloys

Presentation of Equivalent Elements for the Transformation of a Ternary Alloy into a Quinary Alloy

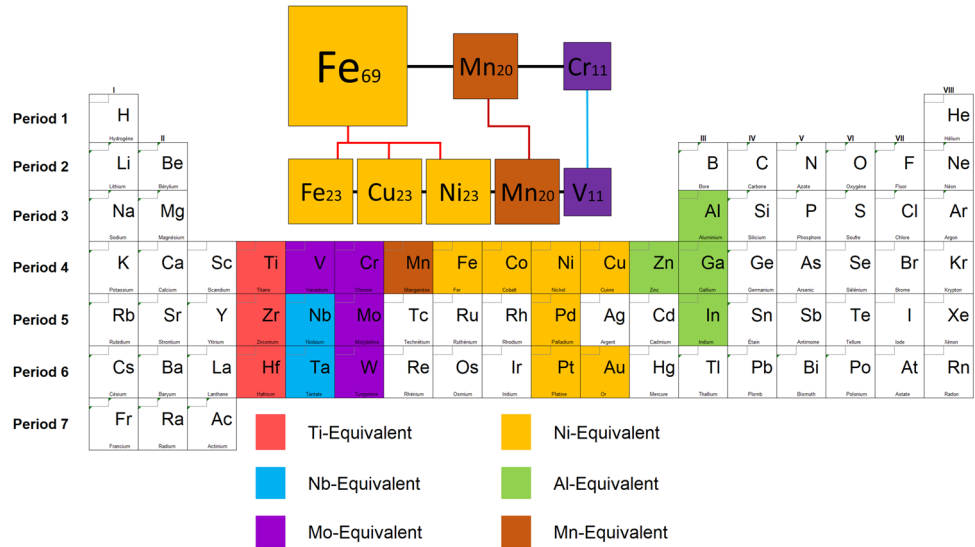
Figure 1 shows the equivalent elements for the six main chemical elements that can constitute alloys with damping properties. In the case of the ternary FeMnCr alloy, Fe can be partially replaced by Co, Ni, and Cu in period 4, Pd in period 5, Pt and Au in period 6. The element Cr can be partially replaced by V in period 4, Mo in period 5, and W in period 6. The element Mn has no other equivalent.

In the design of a new damping alloy that can be used over a wide temperature range, some element pairs such as Ti–Ni and Cu–Al must naturally be avoided to limit the appearance of the thermal memory effect. This temperature dependency of the damping effect is highlighted by the studies of Piedboeuf et al. [25], He et al. [26], Yin et al. [27], and Toker et al. [28] in the titanium and copper-based SMAs. On the other hand, FeNi-X SMAs (where X is Cu, Cr, Mn, Mo) are known to exhibit a non-thermal martensitic phase transformation [40–42].

Presentation of the HEA Criteria

According to the work by Zhou et al. [2], the high entropy of an alloy is only operative if the mixing entropy value, ΔS_{mix} , is greater than $1.5R$, where R is the gas constant ($R = 8.314 \text{ J K}^{-1} \text{ mol}^{-1}$). In addition, Zhou et al. also show the content of different elements is between 5 and 35 at.%. Several criteria are used in high entropy alloy metallurgy to increase the predictability of the single-phase solid solution formation. These criteria are based on the work of Hume-Rothery [43]. The first criterion is the term ΔH_{mix} , which represents the average of the binary mixing enthalpies obtained between each of the pairs of elements in the alloy. An enthalpy close to zero or negative in the Gibbs equation stabilizes the formation of solid solutions. The second criterion concerns the difference in atomic radius for each element constituting the alloy. It should be noted that the integration of the different atoms in the HEA crystalline lattice is easier when the difference between the atomic radii (R_a) is the smallest. Thus, the solubility of the different elements is increased because the distortion of the crystalline lattice is reduced. The third criterion is the difference in Pauling electronegativity between the “ n ” elements of the alloy. The fourth criterion is the average valence electron concentration (VEC). The third and fourth criteria are used to predict the type of crystallographic structure generated by the development of a high entropy

Fig. 1 Periodic elements table showing the different equivalent elements for those commonly used in damping alloys composition



alloy [44–46]. Based on the work of Cantor et al. [1], the elements of transformation of the ternary alloy into a quinary alloy must all belong to the same period. In our case, only elements from period 4 fit this requirement and were considered in the following to define the composition of the designed alloys. Hence, the different criteria established for obtaining a HEA [4, 44, 46], such as enthalpy, atomic radius difference, the difference of electronegativity between all the constitutive elements and the average of the valence electrons (VEC) are optimized.

Multi-criteria Validation of the Alloy Compositions

The enthalpy criterion is determined based on the enthalpy of all the element pairs contained in the quinary alloy. As illustrated in Table 1, element pairs of period 4 (Ti–Ni, Ti–Fe, Ti–Co, Ti–Zn, V–Co, V–Ni, Cu–Cr, Cu–Fe) have to be avoided since their combination enthalpy value is not ranging between -10 and 10 kJ mol⁻¹.

From the criteria, the chemical equivalent composition was set as Cu_xFe_yNi_zMn₂₀V₁₁; the pseudo-ternary graph Cu_xFe_yNi_zMn₂₀V₁₁ shown in Fig. 2 highlights the five

chosen alloys. In the remainder of the present paper, the alloys will be named as follows:

- A_{0–69} for the ternary Fe₆₉Mn₂₀V₁₁ alloy,
- A_{23–23} for the quinary Cu₂₃Fe₂₃Ni₂₃Mn₂₀V₁₁ alloy,
- A_{29–11} for the quinary Cu₂₉Fe₁₁Ni₂₉Mn₂₀V₁₁ alloy,
- A_{29–29} for the quinary Cu₂₉Fe₂₉Ni₁₁Mn₂₀V₁₁ alloy, and
- A_{11–29} for the quinary Cu₁₁Fe₂₉Ni₂₉Mn₂₀V₁₁ alloy.

Table 2 compares the parameters considered for high entropy for damping HEA of the literature. The CrMnFe–CoNi alloy studied by Cantor et al. [1] perfectly fits the different criteria for obtaining a single solid solution. It is worth mentioning that only the four high entropy alloys proposed in this study meet the five selected criteria unlike the other four alloys in the literature [14, 35–38].

Experimental Procedure

Manufacturing Stages and Heat Treatments

The five alloys A_{0–69} to A_{11–29} were produced using raw materials with a purity greater than 99.9 wt% provided by

Table 1 Chemical mixing enthalpies (kJ mol⁻¹) for atomic pairs among them the alloying elements of the fourth period of the periodic table of elements

	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn
Ti		-2	-7	-8	-17	-28	-35	-9	-15
V	-2		-2	-1	-7	-14	-18	5	-2
Cr	-7	-2		2	-1	-4	-7	12	5
Mn	-8	-1	2		0	-5	-8	4	-6
Fe	-17	-7	-1	0		-1	-2	13	4
Co	-28	-14	-4	-5	-1		0	6	-5
Ni	-35	-18	-7	-8	-2	0		4	-9
Cu	-9	5	12	4	13	6	4		1
Zn	-15	-2	5	-6	4	-5	-9	1	

Fig. 2 $\text{Cu}_x\text{Fe}_y\text{Ni}_z\text{Mn}_{20}\text{V}_{11}$ pseudo-ternary diagram

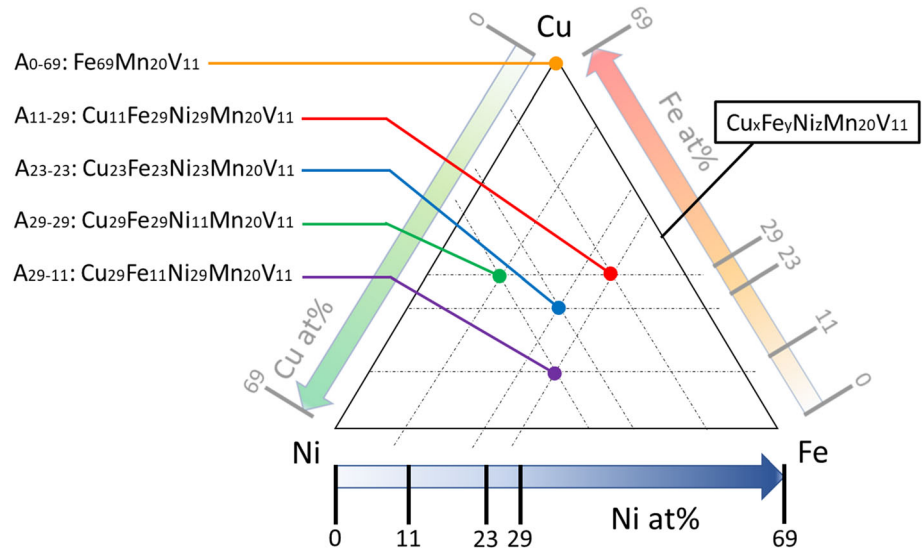


Table 2 Criteria for microstructures formation, ΔS_{mix} and ΔH_{mix} correspond to mixing entropy and enthalpy, respectively, δ corresponds to the relative radii difference, EN means electronegativity, whereas VEC stands for valence electron concentration

Alloys			ΔS_{mix} (J/K/mol)	ΔH_{mix} (kJ/mol)	δ (%)	EN Pauling	VEC
			$x > 1.5$	$x \in [-10; 10]$	$x \in [0; 6.6]$	$x \approx 0$	$x \in [6.87; 7.99]$
CrMnFeCoNi	Cantor et al.	[1]	1.61 <input checked="" type="checkbox"/>	-4.16 <input checked="" type="checkbox"/>	3.27 <input checked="" type="checkbox"/>	0.14 <input checked="" type="checkbox"/>	8.00 <input checked="" type="checkbox"/>
Fe ₆₉ Mn ₂₀ V ₁₁	Present work	A0-69	0.82 <input checked="" type="checkbox"/>	-2.21 <input checked="" type="checkbox"/>	3.60 <input checked="" type="checkbox"/>	0.12 <input checked="" type="checkbox"/>	7.47 <input checked="" type="checkbox"/>
Cu ₂₃ Fe ₂₃ Ni ₂₃ Mn ₂₀ V ₁₁	Present work	A23-23	1.58 <input checked="" type="checkbox"/>	0.33 <input checked="" type="checkbox"/>	3.26 <input checked="" type="checkbox"/>	0.14 <input checked="" type="checkbox"/>	8.62 <input checked="" type="checkbox"/>
Cu ₂₉ Fe ₁₁ Ni ₂₉ Mn ₂₀ V ₁₁	Present work	A11-29	1.52 <input checked="" type="checkbox"/>	-0.26 <input checked="" type="checkbox"/>	3.13 <input checked="" type="checkbox"/>	0.15 <input checked="" type="checkbox"/>	8.92 <input checked="" type="checkbox"/>
Cu ₂₉ Fe ₂₉ Ni ₁₁ Mn ₂₀ V ₁₁	Present work	A29-29	1.52 <input checked="" type="checkbox"/>	3.64 <input checked="" type="checkbox"/>	3.20 <input checked="" type="checkbox"/>	0.14 <input checked="" type="checkbox"/>	8.56 <input checked="" type="checkbox"/>
Cu ₁₁ Fe ₂₉ Ni ₂₉ Mn ₂₀ V ₁₁	Present work	A29-11	1.52 <input checked="" type="checkbox"/>	-3.04 <input checked="" type="checkbox"/>	3.40 <input checked="" type="checkbox"/>	0.14 <input checked="" type="checkbox"/>	8.38 <input checked="" type="checkbox"/>
Fe ₅₀ Mn ₂₀ Co ₁₅ Cr ₁₅	Wang et al.	[14]	1.23 <input checked="" type="checkbox"/>	-1.32 <input checked="" type="checkbox"/>	3.35 <input checked="" type="checkbox"/>	0.12 <input checked="" type="checkbox"/>	7.65 <input checked="" type="checkbox"/>
Al _{0.25} CoCrFeNi	Ma et al.	[35]	1.53 <input checked="" type="checkbox"/>	-6.75 <input checked="" type="checkbox"/>	3.49 <input checked="" type="checkbox"/>	0.11 <input checked="" type="checkbox"/>	7.93 <input checked="" type="checkbox"/>
CrMn _{0.5} FeVCu _{0.1}	Song et al.	[36]	1.44 <input checked="" type="checkbox"/>	-1.94 <input checked="" type="checkbox"/>	3.25 <input checked="" type="checkbox"/>	0.10 <input checked="" type="checkbox"/>	6.56 <input checked="" type="checkbox"/>
Fe _{85.4} Cr ₁₁ Mo _{2.5} Zr _{0.1} Ni ₁	Chen K. et al.	[37]	0.52 <input checked="" type="checkbox"/>	-0.74 <input checked="" type="checkbox"/>	1.78 <input checked="" type="checkbox"/>	0.08 <input checked="" type="checkbox"/>	7.74 <input checked="" type="checkbox"/>
Fe ₆₀ Mn ₂₀ Co ₅ Cr ₁₅	Chen Q. et al.	[38]	1.06 <input checked="" type="checkbox"/>	-0.56 <input checked="" type="checkbox"/>	3.39 <input checked="" type="checkbox"/>	0.12 <input checked="" type="checkbox"/>	7.55 <input checked="" type="checkbox"/>

Criterion not validated
 Criterion validated

Goodfellow company. The alloys were elaborated in a cold crucible, equipped with a pyrometer that measures the melting temperature (T_m) of the alloy. Before heating, the protective atmosphere of the furnace was flushed five times with ultra-pure Argon and a vacuum pump. Melting was performed with a pressure of 900 mBar of ultra-pure Argon. Five melting cycles were performed to ensure the homogeneity of the different alloys. The ingots were then cast in a cylindrical mold of 16 mm diameter and 30 mm height. Three metallurgical states were considered for further characterization: the as-cast state (AC) corresponds to the state just after melting, the ST state is after solution heat treatment (900 °C 30 min) followed by severe water quenching, and a cold-worked state (CW) was obtained by cold rolling. Three industrial alloys were also considered for

damping tests: the 1050A aluminum alloy, the NiTi alloy, and the CuAlNi alloy were in an as-received state (AR state).

Damping Behavior Characterization

The drop weight test is commonly used to determine the hardness and/or the damping behavior of a material [47–51]. In the present work, a calibrated metal ball (bearing steel, diameter 5 mm) was falling in a transparent quartz tube (1 m long). The ball was hitting a cylindrical sample to test its damping behavior. The test bench (Fig. 3) consisted of a heavy treated steel base with a cylindrical socket. The cylindrical sample (diameter 10 mm) of the tested alloy was positioned into the socket before testing. A manual trigger was allowing releasing the ball to launch the

test (Fig. 3c). A mechanical blocker was stopping the ball before the second bounce, which prevented multiple impacts on the surface of the tested sample. During the entire test, a high-speed camera (Fig. 3b) recorded the ball's fall, the impact and the first rebound. The steel base and the sample could be heated or cooled to perform temperature rebound tests from -40 to 200 °C (Fig. 3d).

The drop weight tests have been performed for different metallurgical states and temperatures of the samples. The rebound height and the dent size after impact were analyzed. The rebound height is directly linked to the damping capability of the alloy. The dent size provides information on the physical phenomena that allow the damping and the dissipation of energy during the impact. The dent diameter and its depth were measured using a profilometer (Tesa-Visio). The tests were repeated 20 times for each temperature to refine the statistics. After each drop, the cylinders were polished to remove the dent and the hardened zone due to the ball impact. Figure 4 shows the three phenomena related to the damping behavior of SMAs. If the tested cylinder is in a 100% martensitic metallurgical state (shape memory effect noted hereafter SME), then the energy absorbed during the impact will be coupled with a residual macroscopic strain of the martensite. If the tested cylinder is in a 100% austenitic metallurgical state (superelastic effect, noted SE) or a two-phase martensite/austenite metallurgical state (pseudoelastic effect or PE), then the energy absorbed during the impact will be combined with a reversible elastic strain of the austenite.

It is commonly admitted that the drop tests may induce experimental scattering in terms of impacted zone and its

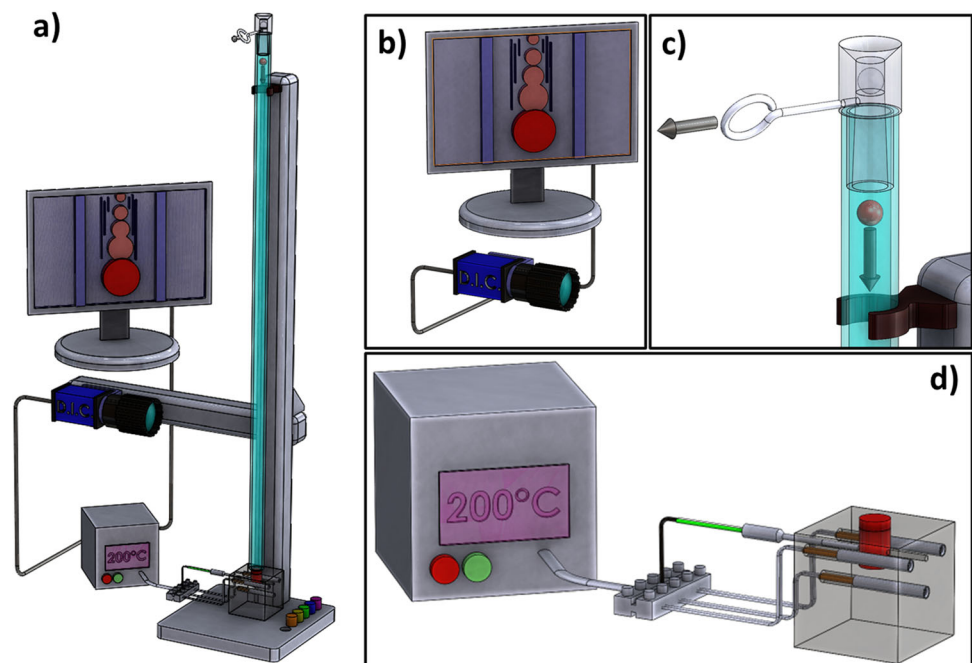
size. On the opposite, the microhardness tests have a good spatial resolution and activate the phase transformation uniformly in the center of the samples and the size of the micro-indentations allows an accurate study by SEM. Vickers microhardness tests were thus performed on an Indentec Zwick ZHV1 machine with a mass of 1 kg and a holding time of 10 s.

Metallurgical Characterization

Microstructure observations were performed using a scanning electron microscope (SEM). The crystallographic structure of phases was identified by X-ray diffraction (XRD) using Cu-K α radiation at room temperature (an accelerating voltage of 40 kV and an intensity of 30 mA); cell parameters were also determined.

A differential scanning calorimeter (DSC) was used to determine the transformation temperatures: the thermal transformation from martensite to austenite during heating was characterized by an endothermic peak and transformation points A_s and A_f (for Austenite Start and Austenite Finish). The reverse transformation from austenite to martensite upon cooling phase was characterized by an exothermic peak and transformation points M_s and M_f (for Martensite Start and Martensite Finish). The area (enthalpy in $J g^{-1}$) and the spacing (hysteresis) of the transformation peaks were also determined. The transformation temperatures were obtained according to ASTM F2004-17 [52] (sample of 40 mg, cooling and heating rate of 10 °C min^{-1}) using a SETARAM DSC 131 calorimeter. Enthalpy changes (ΔH) were obtained using the SETSOFT software.

Fig. 3 An overview of the drop weight set-up: **a** General view of the drop column, **b** Monitoring of the DIC, **c** Manual trigger, **d** In red, cylindrical sample positioned in the instrumented heating steel base (color figure online)



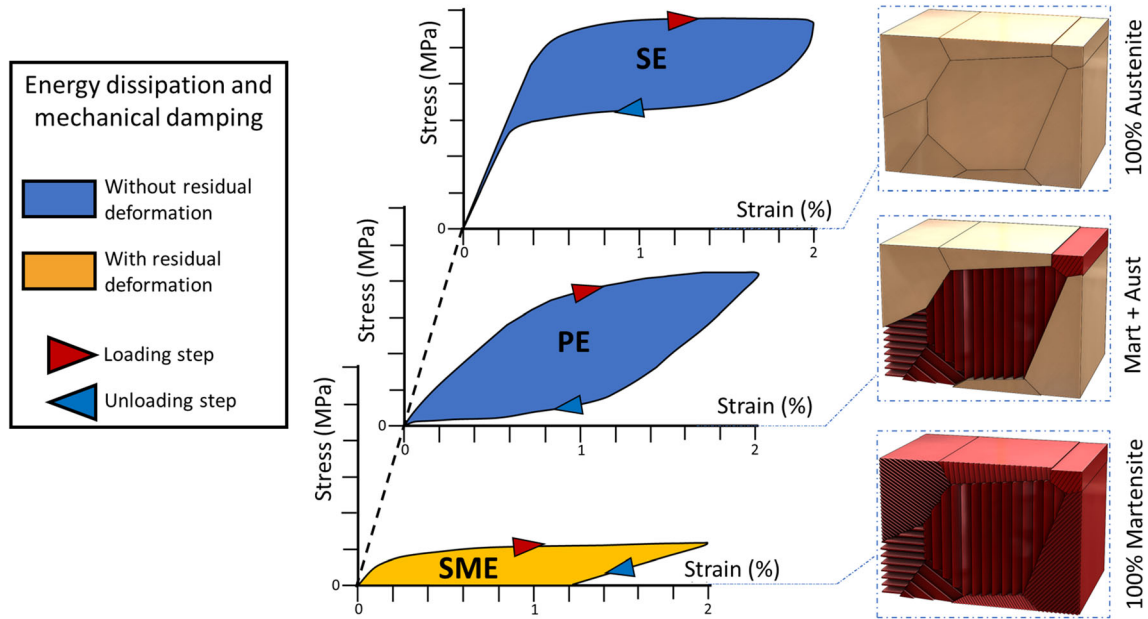


Fig. 4 Damping behavior of SMAs: superelasticity (SE) and pseudoelasticity (PE) are not coupled with residual strain, shape memory effect (SME) is coupled with residual strain

Experimental Results

Analysis of the Stress-Induced Phase Transformation at Room Temperature

First, drop weight tests were performed at room temperature on all the alloys considered in the ST state. XRD analysis was performed on the initial state and inside the dent after the impact (Fig. 5). Diffractograms have been indexed considering FCC or BCC structures. The ternary A_{0-69} alloy was mainly martensitic; weak peaks of austenite could be observed at the initial state but were not anymore present in the dent. The austenite cell parameter was equal to 0.353 nm. The A_{23-23} and A_{29-29} alloy diffractograms show two double peaks; they could have been indexed as two different austenite phases differing by their cell parameters, namely: $a = 0.359$ nm and $a = 0.352$ nm. No martensite was visible before nor after the impact as no peak was detected around 65° . The presence of two phases was confirmed by SEM observations (Fig. 6b, d): dendritic and globular structures were observed in the alloys. For the A_{29-11} alloy, diffraction peaks spread over a large 2θ -angle range, meaning that the two austenite phases were also present in this alloy; the SEM microstructure (Fig. 6c) shows a fine dendritic structure with two different phases. Only the A_{29-29} alloy was exhibiting one single-phase corresponding to the austenite with a cell parameter $a = 0.352$ nm; this was consistent with the SEM observation (Fig. 6). The cell parameter values are in agreement with the work of Yağcı et al. [53] on two ternary low entropy alloys: a martensitic FeMnV alloy with

$\Delta S_{\text{mix}} = 0.54R$ and an austenitic FeMnCu alloy with $\Delta S_{\text{mix}} = 0.58R$. They are also in agreement with the results of Choi et al. [54] who studied $V_{10}Cr_{10}Fe_{45}Co_{10}Ni_{35}$, a medium entropy alloy with $\Delta S_{\text{mix}} = 1.39R$. This composition is close to the FeMnV-like one. Nevertheless, they had only one austenite phase ($a = 0.359$ nm).

As no martensite could be detected in the HE-SMAs after the drop test, Vickers indentation tests have been performed to determine if stress-induced martensite could be induced by one austenite or another.

Figure 6 shows SEM images of indentation. As can be seen, the indentation shows similar sizes whatever the alloy. The alloy's hardness was obtained by performing 10 Vickers tests; average values are reported in Table 3. The results of the hardness tests on NiTi, CuAlNi, and 1050A alloys were consistent with the literature [50, 55, 56]. The hardness of the ternary alloy A_{0-69} was similar to that of NiTi alloy and CuAlNi alloy (about 250 HV1). Except for the hardness of the 1050A alloy, one can claim that the A_{23-23} alloy has the lowest hardness (176 HV1) while the three other HE-SMAs have an equivalent hardness (around 185 HV1).

SEM observations reveal that the Vickers indentations in A_{0-69} , A_{23-23} , A_{29-11} , A_{29-29} , and A_{11-29} alloys (in the ST state) induce inelastic strain. Figure 7a shows the residual indentation with a mass of 1 kg. The size of the affected zone by this transformation varies according to the alloys and their microstructure. A mapping of this affected zone is shown in Fig. 7b.

The area where the inelastic deformation has occurred varies between the alloys. The smallest zone was observed

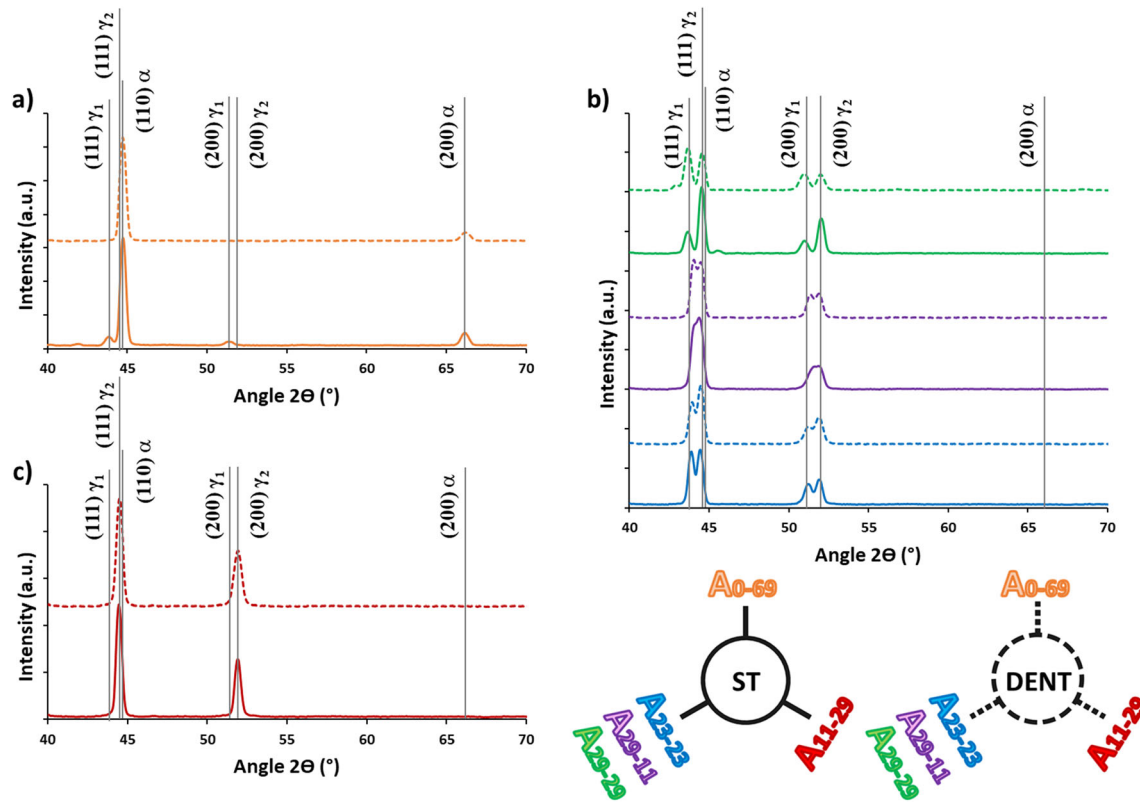


Fig. 5 Diffractograms of the alloys: **a** A_{0-69} ternary alloy, **b** A_{23-23} , A_{29-29} and A_{29-11} quinary alloys and **c** A_{11-29} quinary alloys. The diffractograms of the metallurgical states are differentiated: in solid

lines are represented the diffractograms of the ST state and in dotted lines are the diffractograms inside the dent after the weight drop test

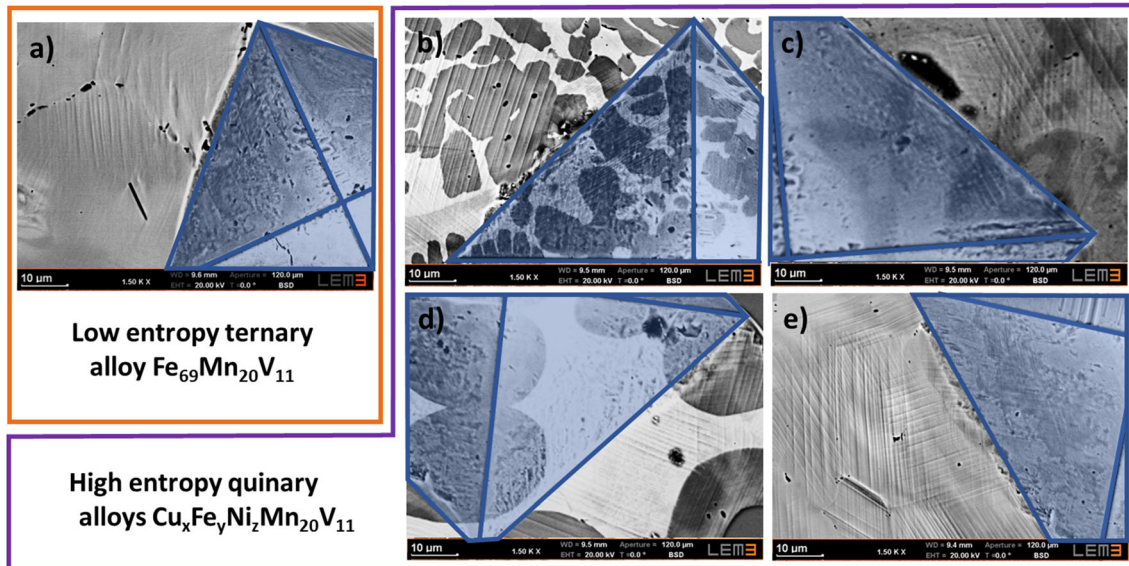


Fig. 6 SEM images of the edges of the indentations for the different alloys studied: **a** ternary single-phase A_{0-69} alloy, **b** A_{23-23} , **c** A_{29-11} , **d** A_{29-29} , **e** single-phase quinary A_{11-29} alloy. The indentation is represented in blue on the five thumbnails (color figure online)

in the ternary A_{0-69} alloy which was martensitic. Inelastic strain by martensite reorientation is thus supposed to take place. The four quinary HE-SMAs were austenitic, so

martensitic transformation was expected near the indentation. The most extended phase transformation was observed on the HE-SMA A_{11-29} ; its initial microstructure

Table 3 Vickers hardness of the different samples studied

Alloys	1050A	NiTi	CuAlNi	A ₀₋₆₉	A ₂₃₋₂₃	A ₂₉₋₁₁	A ₂₉₋₂₉	A ₁₁₋₂₉
Hardness HV1	37	231	215	253	176	192	189	185

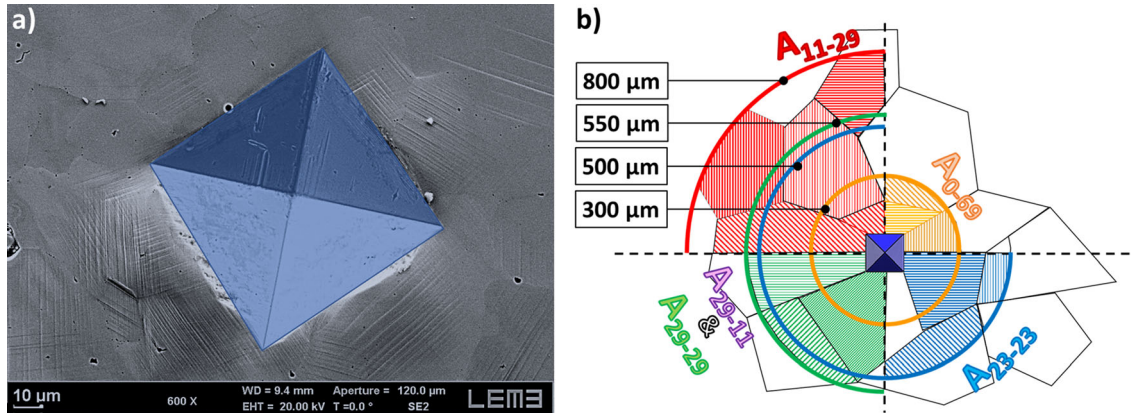


Fig. 7 Observation of the inelastic deformation induced during a hardness test: **a** on the edge of an imprint on alloy A₁₁₋₂₉, **b** cartographic representation of the zone affected by the phase transformation of the different alloys. The indentation is shown in blue (color figure online)

consisted of one single austenitic phase. Martensite was observed until a distance of 800 μm from the indentation center, i.e., 8 times the diagonal size of the indentation; it is 2.7 times larger than the A₀₋₆₉ alloy. The multiphase alloys A₂₃₋₂₃, A₂₉₋₁₁, and A₂₉₋₂₉ had an equivalent phase transformation affected zone, spreading until 500–550 μm far from the center of the indentation. The damping effect of the SMAs is intrinsically linked to the phase transformation. If the phase transformation spreads over a larger distance in the A₁₁₋₂₉ alloy, its damping effect should be higher than that observed for the other HE-SMAs and the ternary alloy.

Thus, a good damping shape memory alloy is characterized by a large hysteresis (about 40 $^{\circ}\text{C}$) and a high enthalpy (above 4 J g^{-1}) [57]. Figure 8 shows the thermograms of studied alloys (except for 1050A) over a range of -60 to 300 $^{\circ}\text{C}$. NiTi and CuAlNi alloys show a good reversible transformation. The ternary alloy A₀₋₆₉ (FeMnV) shows a reversible phase transformation upon heating and cooling but with low enthalpy of 1 J g^{-1} (10 times less than the NiTi alloy). For A₂₃₋₂₃ and A₂₉₋₂₉ alloys in the ST state, the curve on cooling is flat while a high-intensity peak (5 J g^{-1} of the order of copper SMAs) is observed on heating. For A₂₉₋₂₉ and A₁₁₋₂₉ alloys in the ST state, no peak corresponding to the transformation could be distinguished on the thermograms during the heating or cooling stages. This absence of significant peaks by DSC is due to an absence or too low thermal martensitic transformation enthalpy for A₂₉₋₁₁ and A₁₁₋₂₉ alloys. These flat thermograms are in agreement with their diffractograms which show only austenitic phases. These flat thermograms are in agreement with their diffractograms

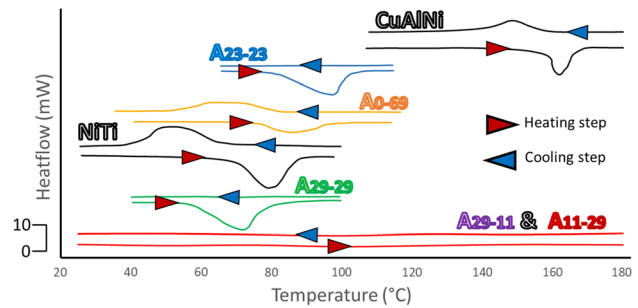


Fig. 8 Thermograms of studied alloys in the ST state

which show only austenitic phases. It should be noted that only the ternary CuAlNi alloy has a phase transformation that occurs above 100 $^{\circ}\text{C}$.

Analysis of the Damping Capacity at Different Temperatures

Figure 9 shows the average height of the first ball bounce on each alloy at different temperatures. During the analysis of the stored images using the digital correlation camera, several types of damping responses occurred. The first one was observed for five alloys: NiTi, CuAlNi, A₀₋₆₉, A₂₃₋₂₃, and A₂₉₋₂₉. The damping behavior of these alloys drops drastically above the A_f transformation temperature: the rebound height was increasing with increasing temperature. This kind of response is the less interesting one for the study. The second kind of response was observed only for the 1050A alloy. The rebound height decreases when increasing the temperature of this alloy. However, the depth and diameter of the dent left after the first rebound

were increasing with temperature due to the high ductility of this nearly pure (or commercially pure) low entropy alloy. These results are in agreement with the literature [58, 59]. This behavior was not relevant to the current study too.

The third type was observed for the alloys A_{29-11} and A_{11-29} . The damping behavior of these two HEAs was evolving slightly but linearly with temperature. These results were in agreement with the thermograms of these two alloys which did not show any phase transformation related to temperature. Except for the 1050A alloy, the A_{11-29} alloy has the highest damping level over the four studied temperatures.

Figure 10 represents the evolution at room temperature of the average hardness, the average diameter of the dent

and the average height of the first bounce for all the studied alloys.

For the 1050A alloy, one notices a correlation between the dent size and the hardness of the sample, which is related to the damping capacity. For instance, the 1050A alloy exhibits the largest dent diameter (360 μm) that corresponds to the smallest measured hardness (37 HV) (Fig. 10). Indeed, as shown in Fig. 9, whatever the temperature, this alloy exhibits the lowest bounce height characterizing hence a high damping capacity. In fact, the damping of a ductile sample occurs through the plasticity where the anelastic energy during the impact is partially dissipated whereas the subsequent part is stored through the irreversible plastic strain [51]. It is worth noticing that this correlation has not been clearly observed for the other seven studied alloys even if they had similar hardness and dent size (Fig. 10).

It is known that for an alloy, lower is the rebound height better is the damping capacity. However, for an SMA, it is suitable to also minimize the impact imprint to avoid other irreversible effects such as plasticity and/or TRIP effects that can modify the overall SMA behavior and may reduce its durability. Thus, the most interesting SMAs, in terms of damping capacity, are those combining a low rebound height with a reduced imprint size.

To this end, a performance indicator has been proposed to reflect an implicit relation between the value of HE-SMAs damping and the rebound height as well as the diameter of the impact imprints. This performance indicator facilitates the data-reduction of the experimental findings and comparison between the HE-SMA's damping capacity. It is expressed as the weighted sum of the normalized rebound height (λ_1) and dent diameter (λ_2). It reads as follows:

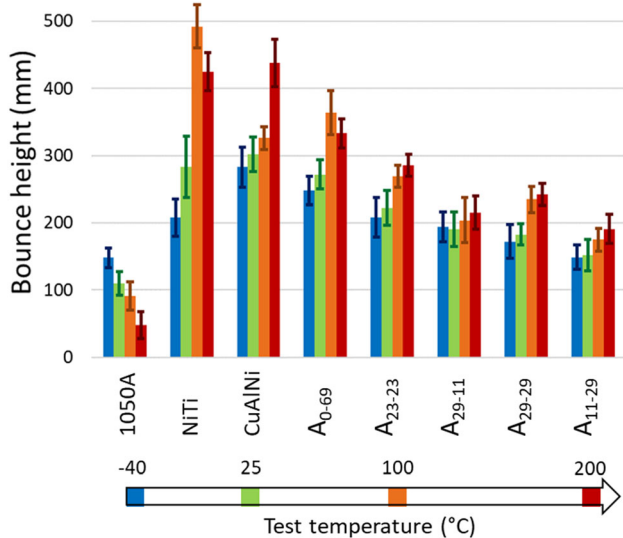
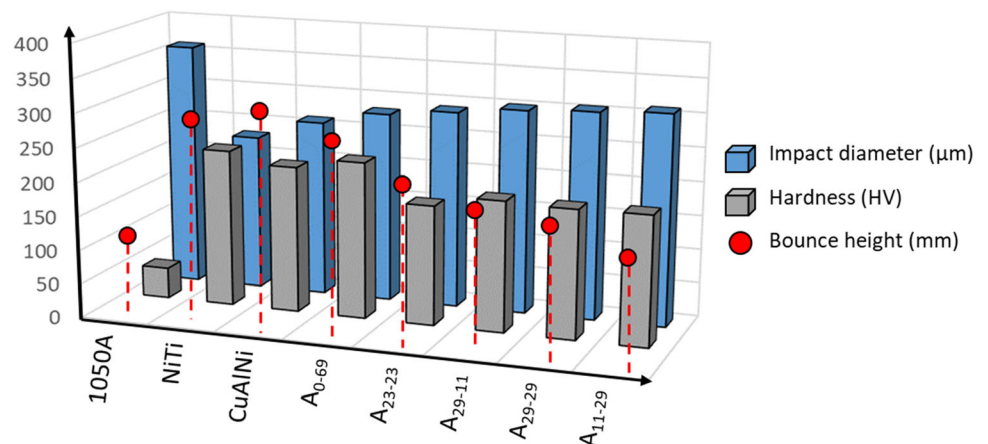


Fig. 9 Representation of the damping effect of the different alloys studied as a function of temperature. The bounce height is inversely proportional to the damping capacity of the alloys

Fig. 10 Graphical representation of the three quantities for each of the studied alloys at room temperature: in red the average height of the ball rebound after impact (mm), in blue the average diameter of the imprint left by the impactor after impact (mm), in grey the average hardness (HV) in the *ST metallurgical state* (color figure online)



Effects of Aging on Damping Properties

$$F = \alpha\lambda_1 + (1 - \alpha)\lambda_2,$$

$$\text{where } \lambda_1 = \frac{h_{\max} - h}{h_{\max} - h_{\min}}$$

$$\lambda_2 = \frac{d_{\max} - d}{d_{\max} - d_{\min}}$$

and α : ponderation factor.

As mentioned above, the most interesting SMAs, in terms of damping capacity, are thus characterized by a performance indicator close to 1.

Figure 11 represents the evolution of the performance indicator with respect to the operating temperature range (-40 °C, 25 °C, 100 °C, 200 °C) for all the alloys. The performance indicator is computer considering an iso-weighting of the rebound height and the dent's imprint diameter. The ponderation factor is then set to $a = 0.5$.

It clearly shows the advantage of using the high entropy damping alloys notably A_{29-11} and A_{29-29} and specifically the HE-SMA, A_{11-29} over a wide temperature range. In fact, by considering a performance indicator ranging between $[0.8-1]$, one can notice that the damping capacity of the A_{11-29} alloy is the most important and remains high for all the operating temperatures compared to the other tested alloys. This is the only alloy composed of the austenite observed first in this study, with a cell parameter of 0.352 nm. So this phase seems very promising for damping properties. In addition, as expected one can confirm that the NiTi alloy exhibits a very good damping capacity from -40 to 25 °C but it loses its damping capacity at relatively high temperatures between 100 and 200 °C.

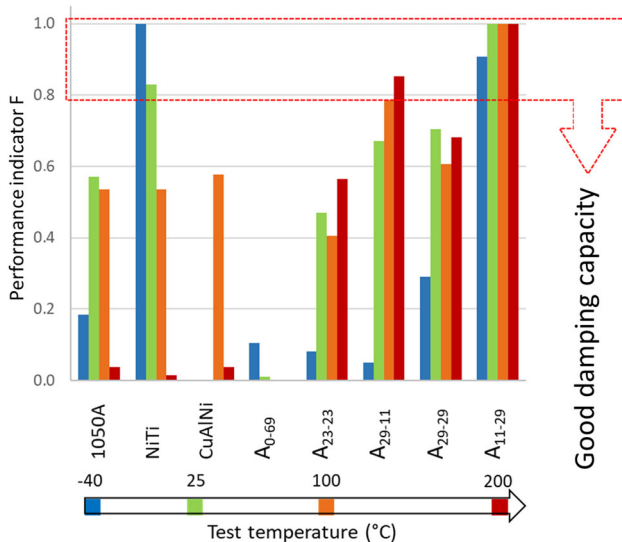


Fig. 11 Performance indicator representation of the different studied alloys for a wide range of operating temperatures

This lack of explicit correlation between hardness and damping effect has been reported in beta-titanium alloys or SMAs [60, 61]. The loss of damping effect at room temperature or during thermal aging treatment has been also studied previously by Zhong et al. [31], Liu et al. [33] and Zhang et al. [62]. In the present study, an accelerated aging has been performed in an oven without a protective atmosphere (800 h at 250 °C) for all the alloys.

After one month of aging (800 h), a preliminary qualitative analysis has been performed. It has been noticed that some samples have altered their color due to the thermal oxidation visible to the naked eye. This was all the more visible on the CuAlNi alloy which showed a dark matte black oxide. The NiTi alloy turned into a golden-brown color. The A_{0-69} alloy turned blue after aging. For the other samples (1050A, A_{23-23} , A_{29-11} , A_{29-29} , and A_{11-29}), no color modification was observed.

A quantitative analysis was carried out using SEM observations realized on the aged samples. Both NiTi and A_{0-69} alloys were subject to precipitation at the grain boundaries, grain growth has occurred in the CuAlNi alloy as reported in the literature [63]. The variation of the damping properties of the alloys after the aging period with respect to those determined on the unaged alloys at room temperature is presented in Table 4.

The 1050A alloy presented a decreased hardness by 24% while the dent diameter was increased by the same proportion in comparison to the unaged state at room temperature. Thus, the aged sample was presenting a larger plastic zone whereas the rebound height decreased by 45%.

The NiTi alloy has developed precipitations during the aging [64, 65]. The precipitates have induced a hardening of the NiTi alloy (21% increase in hardness) but they also have limited the reversible martensitic phase transformation. The martensite was then retained after impact and the NiTi alloy behavior was similar to a ductile alloy, such as the 1050A where the damping capacity is induced by plasticity. This is confirmed by the 60% increase in dent size.

In the CuAlNi alloys, cracks located at the grain boundaries were visible after the impact: the weight drop test has entailed grain de-cohesion. Moreover, its hardness has increased by 30%. This increase was not compatible with grain growth as reported in [66, 67]. Precipitates have been observed by SEM analysis at the grain boundaries; this has induced brittleness and a cracks phenomenon in these areas [68].

The low entropy ternary alloy A_{0-69} has displayed a hardening by precipitation (40%) and the rebounds were 14% higher than those measured on unaged samples of this alloy. The average dent size was decreased by the same

Table 4 Property variations at room temperature after aging

Alloys	Property variations after aging (%)		
	Hardness	Bounce height	Impact diameter
1050A	– 24.4	– 45.5	25.0
NiTi	21.2	7.1	60.9
CuAlNi	33.0	20.0	7.7
A ₀₋₆₉	40.3	14.8	– 14.3
A ₂₃₋₂₃	25.0	22.7	– 3.4
A ₂₉₋₁₁	9.4	5.3	– 1.7
A ₂₉₋₂₉	11.1	11.1	1.6
A ₁₁₋₂₉	8.1	3.3	0.3

Bold data indicates the alloys whose characteristics change the most after the aging heat treatment either in positive or negative for the 3 data studied.

amount. The damping capacities of A₂₃₋₂₃ (strongly dendritic) and A₂₉₋₂₉ (globular) alloys were decreased by the aging heat treatment. Indeed, the rebound heights, as well as the hardness of these two alloys, were higher than those of unaged samples and in the same proportions but their impact sizes were remaining comparable to the unaged configuration. A₂₉₋₁₁ (low dendritic) and A₁₁₋₂₉ (single-phase) alloys were the alloys whose damping capacities were less degraded by the aging heat treatment as shown in Table 4. Indeed, the evolution due to the aging was limited and did not exceed 9% for the hardness, 5% for rebound and almost no modification of the dent size. This confirms again that the new austenite phase is very stable over a large range of temperatures even for long-duration treatment.

Cold Workability

To be used in industrial applications, alloys must have good formability. Also, the cold formability of studied ST alloys A₀₋₆₉ to A₁₁₋₂₉ was carried out by cold rolling at room temperature using a jeweler's roller mill. Square cross-section samples ($7 \times 7 \times 6 \text{ mm}^3$) were cut by machining from the ingots and rectified to have the same initial cross-section for each sample. Each rolling phase consisted of several successive steps where the sample was placed between the rollers of the rolling mill to reach a strain level of 10%. After each step, the hardness of the sample was measured. The sample was kept for the next steps if and only if no crack was observed. The formability analysis was stopped for a sample as soon as it was containing detectable cracks or when its thickness was equal to 1 mm, corresponding to a thickness reduction of 80%. The results are reported in Fig. 12; crosses indicate the appearance of cracks. The alloys show different ductility

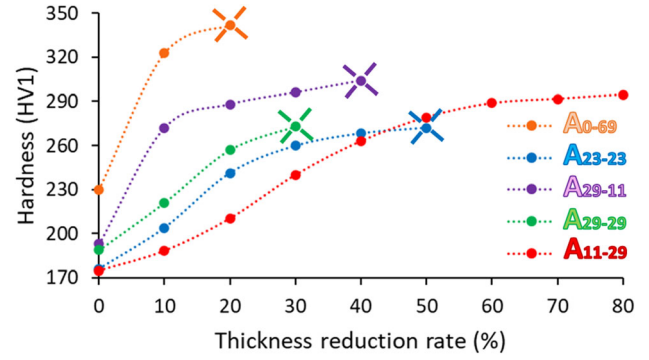


Fig. 12 Analysis of the ductility and the evolution of the hardness of the alloys during the cold rolling phases of A₀₋₆₉ to A₁₁₋₂₉ alloys in the ST state. The A₁₁₋₂₉ alloy exhibit a capacity for structural hardening as well as an important capacity for the cold forming process

levels and cold working hardening abilities. The ternary A₀₋₆₉ alloy had the highest hardness, even higher than the CuAlNi and NiTi alloys; its formability was limited to a 20% reduction rate. It could be due to its martensitic structure. The quinary A₁₁₋₂₉ alloy, which is a single-phase alloy, had the highest ductility. It was the unique alloy that did not show crack, even at 80% reduction. As in previous studies, the ductility of some single-phase HEAs and HE-SMAs has been shown by Lohmuller et al. [69] and Peltier et al. [70] whose chemical compositions of alloys validate the five criteria related to high entropy in Table 2 (i.e., ΔS_{mix} , ΔH_{mix} , δ , EN and VEC).

Concluding Remarks

In this study, CuFeMnNiV-based HE-SMAs show excellent damping performance, with a damping value that represents two times that of nitinol in the case of the four studied operating temperatures ranging between [– 40 to 200 °C].

The successful development of new HEAs can be enhanced by the use of the following parameters: configurational entropy ΔS_{mix} , enthalpy of mixing ΔH_{mix} , atomic size difference δ , Pauling electronegativity and valence electron concentration VEC.

The use of the equivalence table allows the transposition of a ternary FeMnCr shape memory alloy into quinary high entropy shape memory alloys: the four manufactured HE SMAs have a martensitic phase transformation. The choice of replacing the Cr element with the V is confirmed by the relevant work of Utegov in 2008 [71], which shows that when Cr is substituted by V, these alloys display higher damping properties.

As reported in the literature, the best performing alloys in terms of damping are magnesium-based. This is particularly the case when the alloys contain iron and nickel.

The cocktail effect characterizing the HEAs allows the mixing of several elements of CuNiFeMnV leading hence to optimized damping capacity for the developed HE-SMAs. As in previous studies [70, 72], the sluggish diffusion due to the high entropy character of the alloys maintains the damping capacity of the HE-SMAs at high temperatures as well as during a long-term aging process.

The variation of the copper, nickel and iron content in the different alloys induced the evolution of the number and the nature of the phases observed in the obtained alloy. The two single-phase alloys (A_{0-69} and A_{11-29}) are those whose enthalpy of mixtures is negative but higher than -10 . The alloys with dendritic phases are the alloys whose mixing enthalpies are close to 0 (A_{23-23} and A_{29-11}). The multi-component alloy A_{29-29} has a score of mixing enthalpies of 3.64.

This study demonstrates that the chemical composition of alloys affects the ductility and suitability for structural hardening by cold forming. Zhang et al. [7] explain that these two characteristics are important in the choice of a damping alloy for industrial application. The austenitic A_{11-29} alloy shows a higher ductility, a better damping behavior and higher thermal stability between -40 and 200 °C than the low entropy alloys FeMnV, NiTi, and CuAlNi.

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