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Effect of electro discharge machining (EDM) on the AISI316L SS white layer microstructure and corrosion resistance

Habib Sidhom · Farhat Ghanem · Tidiane Amadou · Gonzalo Gonzalez · Chedly Braham

Abstract The localised corrosion resistance of austenitic stainless steels is strongly influenced by the quality of finished surface. EDM machining induces substantial changes by the high thermal gradients generated by electric sparks. Experimental techniques such as roughness measurement, scanning electron microscopy (SEM), energy dispersive microanalysis (EDX) and X-ray diffraction technique, reveal microgeometrical, microstructural, chemical and mechanical changes. These changes lead to white and heat-affected layers with a depth less than 100 μm . The white layer is a melted material characterised by dendritic structure and constituted by austenite, chromium carbide and ε -carbide. The heat-affected layer is characterised by very large grain size comparatively to the bulk material. Electrochemical test coupled with metallographic examinations using SEM reveals a weakening of the resistance to pitting and intergranular corrosion comparatively to diamond polished surface. This weakening is correlated to differences in structure and chemical composition of white layer. Susceptibility to stress corrosion cracking has been

attributed to the field of tensile residual stresses resulting from thermal effects. The removal of the white layer material by polishing or wire brushing restores the corrosion resistance of the AISI316L SS.

Keywords EDM · Corrosion · Surface · Microstructure · White layer · Residual stress

1 Introduction

The complex part shapes in metallic carbides or high strength steels are usually manufactured by high energy density processes like electron beam, laser or electric discharge machining (EDM) [1–3]. The main advantage of these processes is the low dimensional dispersion comparatively to conventional machining process. This high accuracy results from the absence of physical contact between the sample and the tool by avoiding completely the effect of mechanical vibrations. The main disadvantages of these processes are the low-speed machining and the detrimental effects of the high temperature gradients that involve significant changes in the near surface layers [1]. The nature and the extent of these changes depend on the specific process parameters, on the microstructural and thermophysical characteristics of the machined material and on the properties of the electrode tool (in case of EDM) [2]. Numerical studies of thermal aspects of the EDM have been developed for different types of thermal loading and different types of machined materials. Shuvra et al. [4] and Ghanem et al. [5] used a heat source disc with a constant radius whereas Schuse et al. [6] and Yadav et al. [7] adopted a heat source Gaussian distribution. All these studies confirm that:

- Under the conditions of EDM, metal is removed by successive electrical sparks that arise between the electrode tool and the immersed sample in the dielectric liquid.

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- Each produced spark acts as a heat source and causes an increase in local temperature ($>3,000^{\circ}\text{C}$) for a very short time (from 2 to 100 μs).

Therefore, material that is heated above the melting temperature is removed and flushed away by the dielectric fluid. In addition, the microstructural and mechanical properties of the remaining layers will undergo changes as reported in the literature [1, 2, 5, 6, 8–10]. Numerous studies have been devoted to the identification of metallurgical transformations on the surface and subsurface layers of different materials machined by this process:

- In the case of hardenable steels, EDM leads to a structure of three layers, a white one, a quenched martensite and a transition layer [5, 8, 9].
- In the case of non-hardenable steels, metallurgical transformations result only in the formation of a recast layer with a dendritic structure [5, 10].

These changes create significant reduction in the endurance limit of EDM components with respect to other machining process [8, 9, 11–17]. Ghanem et al. reported a reduction by 35 % in the endurance limit of EDM machined X160CrMoV12 steel components compared to parts obtained by milling process [8, 9]. Grosh [13] has indicated that EDM reduces the endurance limit of the EN X210Cr12A tool steel by 50 % of its value evaluated when turning is used. Tadao and Takeo [14, 15] have shown that EDM lowers by 60 % the endurance limit of the cemented carbide D20 comparatively to polishing process. When EDM was used instead of laser cutting, Kisuke and Katsuji [16] have noticed a reduction by 13 % of the endurance limit of carbon steel (0.45 %C). Hocheng et al. [18] reported an improvement in the wear resistance of EDM machined SiC/A parts, compared to grinding.

The EDM surface corrosion resistance has encountered little interest [19–23] despite of its importance for many applications of metallic components manufactured by this process and used in more or less aggressive environments typically in aerospace, nuclear power plant and biomedical dental implant industries [1–3, 24–28]. Furthermore, the available data on the corrosion resistance of finished surface by EDM is incomplete and sometimes controversial to provide any meaningful conclusion. Indeed, Yan et al. [20] show that surfaces of an Al–Zn–Mg aluminium alloy machined by EDM have greater corrosion resistance, evaluated by weight loss in a 10 % solution of NaOH, compared to surfaces finished by grinding. The authors attribute this enhancement to the role of the amorphous layer formed by EDM in improving the resistance to uniform corrosion of these alloys. They also show that corrosion resistance is significantly improved when defects such as micropores, cracks and discharge craters in the white layer are removed by a special EDM electrode and burnishing by ZrO_2 balls. However, Uno et al. [21] show that

surfaces of low-alloyed steel C (0.15 %), Ni (3 %), Al (1 %) and Cu (1 %), machined by electron beam with large irradiated area have greater resistance to corrosion than surfaces obtained by EDM or by grinding. Authors explain this result by the absence of the white layer and low surface roughness obtained by an electron beam at optimised operating conditions. Based on this result, Tsai et al. [22] developed a sintered electrode tools made in Cu–Cr 80–20 % composite to improve the uniform corrosion resistance of the AISI 1045 EDM surfaces. The authors show an improvement of steel corrosion resistance by reducing the thickness of the white layer and the crack density and by favouring the migration of chromium from the electrode tool to the surface of the work piece. However, Ntasi et al. [23] demonstrated that EDM machining compromise the corrosion resistance of Cr–Co alloys and grade II CpTi implant dental materials. The authors attributed the inferior corrosion resistance of EDM surface, compared to the conventional machining surface, to the role of different elemental composition and surface texture in the occurrence of galvanic phenomenon and an increase of surface vulnerability to pitting and crevice corrosion. Obara et al. [19] have identified specific phenomena of corrosion in the case of the electrical discharge machining of WC–Co composite, by selective dissolution of cobalt in water used as a dielectric fluid.

Kathuria et al. [28] reveal, on the basis of accumulation of wear debris around the implant, the existence of real risks of emission of toxic elements in the human body in the form of Cr^{6+} , Mo^{6+} , Fe^{3+} , Co^{2+} and Ni^{2+} ions, by corrosion in vitro physiological fluid of AISI316L SS used for fabrication of expansion cardiovascular stents by micro EDM or by laser beam. This suggests the need of understanding and assessment of the white layer role on the localised corrosion resistance of AISI316L SS, proposed by this study. The approach used consists to compare the corrosion resistance of surfaces generated by EDM, under different machining conditions, to diamond polished ones. The results will be discussed on the basis of microgeometrical, microstructural and mechanical properties of surface layers. Scanning electron microscopy (SEM) observation, energy dispersive microanalysis (EDX) analysis and X-ray diffraction will be used to investigate microstructure and phase composition of white layer. The resistance to pitting corrosion initiation and propagation, intergranular corrosion (IGC), and residual stress corrosion cracking (RSCC), will be evaluated by an electrochemical tests coupled with SEM examination of surface before and after corrosion testing.

2 Material and machining conditions

This study focuses on the AISI316L SS austenitic stainless steel that is known for its good localised corrosion resistance. This material has been selected to produce some biomedical components by EDM machining that could be

Table 1 Chemical composition of AISI316L SS

Contents (wt%)								
C	Si	Mn	P	S	Cr	Ni	Mo	Fe
0.017	0.56	1.21	0.025	0.001	17.6	11.1	2.16	Balance

sensitive to corrosion in vitro physiological fluid. The chemical composition of this material is reported in Table 1. The as-received microstructural, mechanical and thermal characteristics are reported in Table 2.

EDM machining is performed on a Languopin SE1000 machine, equipped with parameter control system that includes voltage, discharge current, duration and frequency of discharge and dielectric flow. The electrode tool is made in graphite. It has a cylindrical shape containing a hole that is used to inject the dielectric liquid in the inter-electrode gap at a constant flow. The dielectric fluid used was paraffin oil.

The machining operating conditions used in this study are reported in Table 3. These parameters are selected to explore the machining conditions for high and low material removal rates depending on the discharge current (i_e). The average gap voltage (\hat{u}_e) is 46 V. The machining conditions going from roughening ($i_e=50$ A) to finishing ($i_e=2$ A) have been chosen in order to produce a large range of sublayer thickness and properties. The reference specimen was conventionally finished by diamond polishing.

3 Testing

3.1 Identification techniques of surface changes

Surface changes resulting from the material–process interactions under EDM conditions were identified by appropriate experimental techniques:

- Roughness measurement for microgeometrical changes.
- SEM coupled with EDX analyses for chemical and microstructural changes and thermal crack network characterization.
- Vickers microhardness measurements were made to characterise the effect of microstructure and chemical changes on the near surface layer properties.
- X-ray diffraction data for EDM induced phase identification were collected with a Siemens D5000 equipped

with Co-K α radiation, 0.178919 nm, at 35 KV and 30 mA. The 2θ scanned range was 35–130°, with a step of 0.05° and 10 s/step. For structural refinements, the FULLPROF.2 k was used as Rietveld software. The background was approximated by a linear interpolation of 39 data points. The peak profile retained for the refinement study was a Thompson–Cox–Hastings pseudo-Voigt.

- X-ray diffraction technique for residual stress measurements using a SET-X diffractometer equipped with a PSD detector, a high voltage power supply and a classical PSI goniometer. The measurement conditions are reported in Table 4.

3.2 Corrosion tests

The corrosion resistance of the machined surfaces is evaluated by electrochemical tests completed by metallographic examinations under SEM of morphological and elemental alteration before and after corrosion testing:

- Pitting corrosion was evaluated by potentio-dynamic cyclic and potentio-kinetic tests performed in synthetic sea water as recommended by ASTM1141-86, using a Voltalab 32 potentiostat. The tests are performed with a potential scanning rate of 2.5 mV/s in the potential scanning range –300, 600 mV/SCE.
- Intergranular corrosion was evaluated by double loop electrochemical potentiokinetic reactivation on the basis of the density of reactivation current (I_r)/density of activation current (I_a) >1 % criterion (Fig. 1). The operating conditions of the tests are reported in Table 5 as recommended in previous work [29].
- RSCC susceptibility was evaluated by SEM observation of the sample surface before and after immersion during 48 h in a 40 % magnesium chloride solution (MgCl₂) heated at 140°C.

4 Results

4.1 EDM surface characterization

The EDM thermal cycles lead to substantial surface changes. The nature and extent of these changes depend

Table 2 Microstructure, mechanical and thermal properties of AISI316L SS

Metallurgical structure	Thermal conductivity λ (W m ⁻¹ °K ⁻¹) at 25°C	Mechanical properties		
		Ultimate tensile stress UTS (MPa)	Elongation A (%)	Hardness (Hv)
Austenite	16	620	35	230

Table 3 EDM conditions

Machining conditions	Finish	Rough
Discharge current, i_e (A)	2–5–12	20–25–30–50
Discharge duration, t_e (μ s)	5	
Pulse interval time, t_o (μ s)	2	
Duty factor (τ) (%)	80	
Average gap voltage, \hat{u}_e (V)	46	
Open-gap voltage, u_i (V)	120	
Polarity	Positive	
Tool for machining	Graphite (Ellor7, intermediate size of grains $\approx 35 \mu$ m)	
Diameter of the tool electrode (mm)	20	
Dielectric liquid	Paraffin oil (dynamic viscosity) $\mu = 2.4 \times 10^{-3}$ Pa.s	
Flow of dielectric injection in inter-electrode space (liter per hour)	2	

on the machining conditions and particularly on the electric energy converted into heat on the sample surface. Such changes cover microgeometrical, surface integrity, microstructural, chemical and mechanical aspects.

4.1.1 Microgeometrical surface changes

The EDM surface shows rough and porous texture with characteristic melting drops with craters assigned to the collapse of plasma columns during machining. Craters are as deeper as the material volume affected by the heat flow increases (Fig. 2). This volume is directly proportional to the discharge current. The total roughness (R_t) increases from 30 to 320 μ m while the discharge current varies from finishing (2 A) to roughening (50 A) machining conditions (Fig. 3).

4.1.2 Surface integrity

SEM investigations of EDM surfaces reveal the presence of defects such as micropores, deposition of resolidified material and thermal cracks as a consequence of the rapid heating and cooling processes (Fig. 4). The crack distribution seems to be a thermal fatigue damage generated by repeated electric sparks. These electric charge–discharge sequences lead to high amplitude of thermal stresses at the outer layers of the machined material (less than 100 μ m), while the bulk of the material is still cold.

The density of thermal cracks increases with discharge current increasing. It varies from 5 to 25 μ m/ μ m² when the discharge current increases from 2 to 50 A. These cracks are called "short" because they do not penetrate deeply in the

Table 4 X-ray diffraction conditions

Target	Mn
Filter	Cr
Current	5 mA
Voltage	20 KV
Type of goniometer	Psi
Young modulus (E)	196,000 MPa
Poisson ratio (ν)	0.29
Wavelength	2.102 Å
Plane hkl	311 γ
Bragg Angle	152.3°
Beam section	$\varnothing=2$ mm
Number of Psi angles	17
Number of Phi angles	2 (0° and 90°)

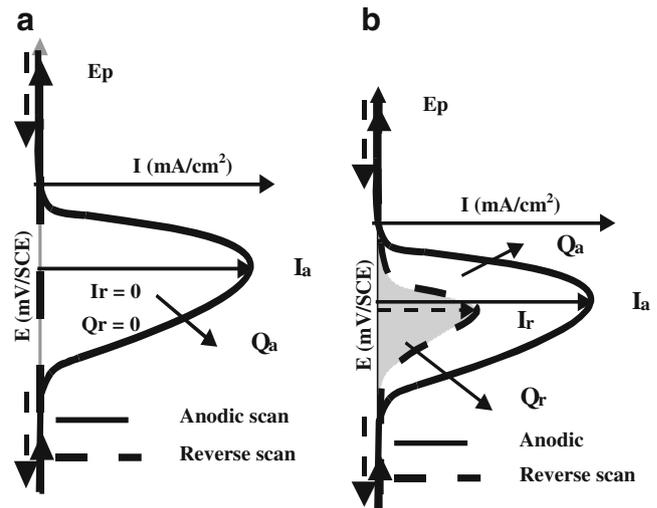
**Fig. 1** Principle of DL-EPR technique and IGC sensitization criterion; **a** solution annealed state and **b** sensitised state

Table 5 Operating conditions of DL-EPR test [29]

Medium	Temperature (°C)	Potential scanning rate(mV/s)	Potential scanning range(mV/ECS)
H ₂ SO ₄ 0.5 M + NH ₄ SCN 0.01 M	25	1	-300↔400

material: they vary from 5 to 40 μm when the discharge current increases from 2 A (finishing EDM) to 50 A (roughening EDM).

4.1.3 Microstructural changes

Optical microscope cross-section examination of EDM machined specimens shows three different layers (Fig. 5):

- White layer corresponding to the melted area with depth irregularities (Fig. 5). The thickness of the white layer varies from 5 to 80 μm when the discharge current increases from 2 to 50 A (Fig. 6). This layer is composed by a series of sublayers marked by solidification fronts during the cooling of the melted material resulting from repeated discharge sparks (Fig. 5).

White layer cross-section backscattering electron image highlights two sublayers with high mean atomic number (white) and low atomic number (black; Fig. 7a):

- Outer layer (white sublayer) with fully dendritic structure likely supposed the recast material formed from unflushed and redeposited EDM removed material (Fig. 7b).
- Inner layer (dark sublayer) with structure characterised by columnar at the interface white layer/bulk material followed by dendrites oriented along the cooling gradient (Fig. 7c).

White layer phases were identified by X-ray diffraction. The X-ray beam penetration depth (an average of 6 μm), allows to irradiate mainly the white layer materials and therefore the identification concerns phases of this zone. Figure 8 revealed three different phases: carbon-enriched

austenite (γ), M₇C₃, and ε-carbides (M₃C; where M=Fe, Cr). Rietveld lattice parameters of these phases are reported in Table 6.

- Heat-affected layer located beneath the white layer, with depth and austenitic grain size irregularities. It corresponds to significantly heated material but not melted during EDM process. The larger grain size results from dynamic recrystallisation at high temperature (Fig. 5).
- Bulk material associated to the core material thermally unaffected austenitic structure (Fig. 5).

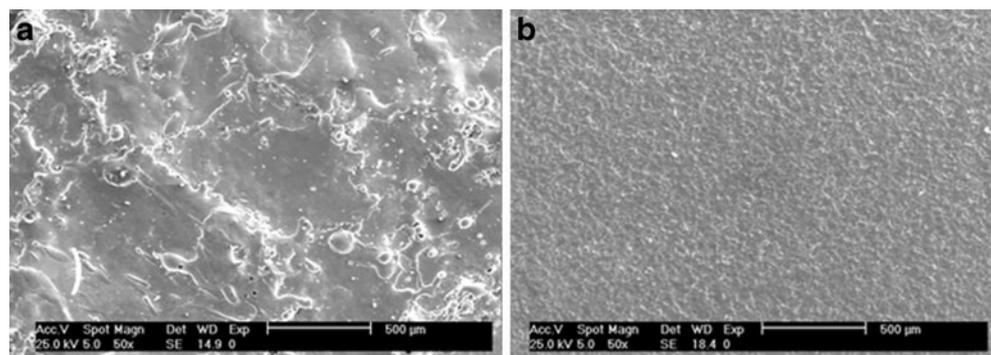
The microhardness measurements reveal a high near surface hardness (700–800 Hv), corresponding to the white layer followed by an abrupt drop to bulk material hardness (200 Hv). The level and the depth of EDM hardening increase with the discharge current increasing (Fig. 9). The high near-surface hardening (~800 HV) is likely assumed to the metallic carbide dispersion in the white layer.

4.1.4 Chemical changes

EDX analysis confirms that the carbon content varies within the white layer, ranging from 3 to 9 wt% at the surface of the workpiece down to 0.017 wt% (formally 0 % considering the accuracy) which is the reference carbon content of the steel. This result is confirmed by EDX measurements realised at the same conditions that conventionally polished surface (Fig. 10).

This suggests that chromium and iron carbides dominate the surface white layer. Then, analysed carbon corresponds to its allied form as suggested in other alloys subjected to EDM machining [10].

Fig. 2 EDM surface texture; **a** roughening conditions ($i_e=50$ A) and **b** finishing conditions ($i_e=2$ A)



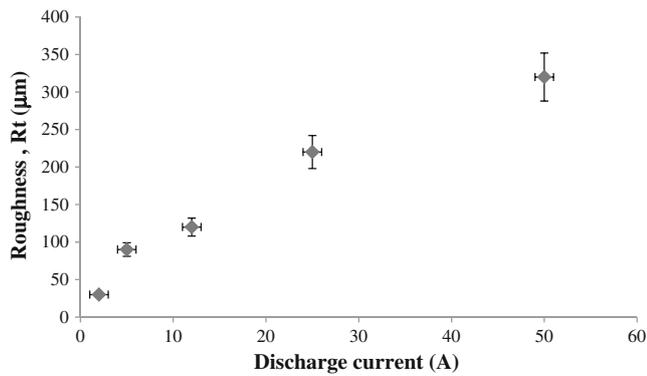


Fig. 3 Effect of EDM conditions on the surface roughness

Back scattering electron image of EDM surface illustrates randomly distributed areas of low mean atomic number (black, zone 1) and high atomic number (white, zone 2). The corresponding quantitative analysis shows that the black areas are rich in chromium (about 25 wt%) while the white layer chromium content is around 15 wt% (Fig. 11). Moreover, X-ray mapping shows that the interdendritic space of redeposited material is Cr- and Mo-depleted zones comparatively to the dendritic axis and arms (Fig. 12).

4.1.5 Mechanical changes

In the EDM-affected layers, a plane isotropic distribution of residual stresses is created with a high gradient at the in depth profile. The maximum values, obtained in sub-layers, increase with increasing the discharge current. The white layer residual stresses are partly released on the near surface (Fig. 13). Such stresses are the consequence of the microstructural changes

Fig. 4 EDM surface cracks and defects

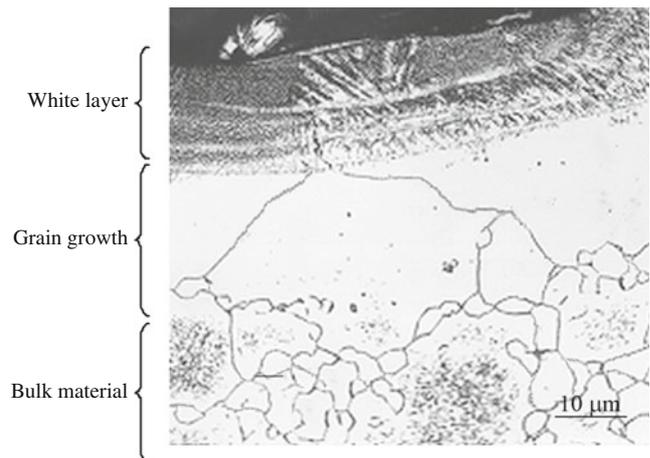
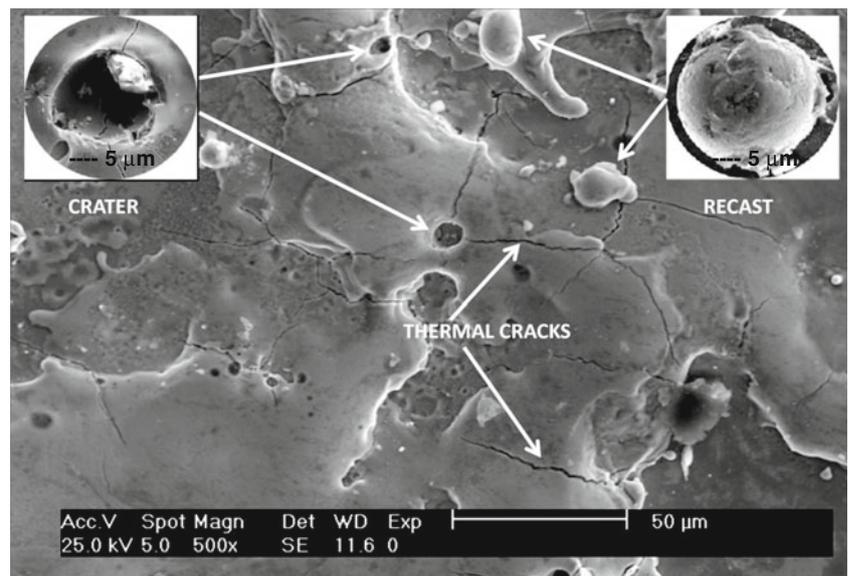


Fig. 5 White and heat-affected layers induced by EDM

and contraction during intense cooling of the surface layers, markedly expanded by the heat load, generated by the EDM process.

4.2 Evaluation of white layer corrosion resistance

The results of cyclic potentiodynamic and potentiostatic tests show that EDM significantly alters the pitting and crevice corrosion resistance of AISI316L SS surfaces in chloride medium (Fig. 14). Moreover, the appearance of I_r during the double-loop electrochemical potentiokinetic reactivation (DL-EPR) tests, performed under appropriate operating conditions [29], is indicative of an IGC sensitivity induced by EDM (Fig. 15). The results of electrochemical tests and micrographic examinations performed on the samples to evaluate the resistance of

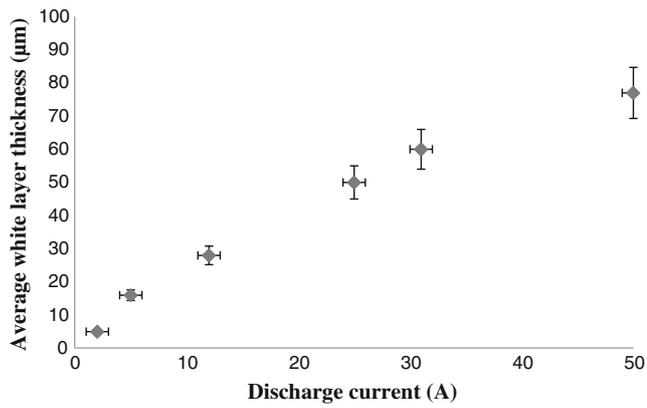


Fig. 6 Effect of EDM discharge current on the average white layer thickness

EDM surfaces to various forms of localised corrosion are compared to those of conventionally polished surface. The corrosion testing data are reported in Table 7 which shows the following comments:

4.2.1 Resistance to pitting initiation and the crevice propagation

The values of pitting critical temperature (PCT, 16–32°C), pitting potential (E_p , 12–230 mV/SCE), and repassivation potential (E_r , -300 to -50 mV/SCE), reported in Table 7 (column D) are lower for EDM surfaces compared to those

Fig. 7 White layer structure; **a** recast and melted sublayers, **b** dendritic structure of recast material and **c** dendritic and columnar structures of melted material

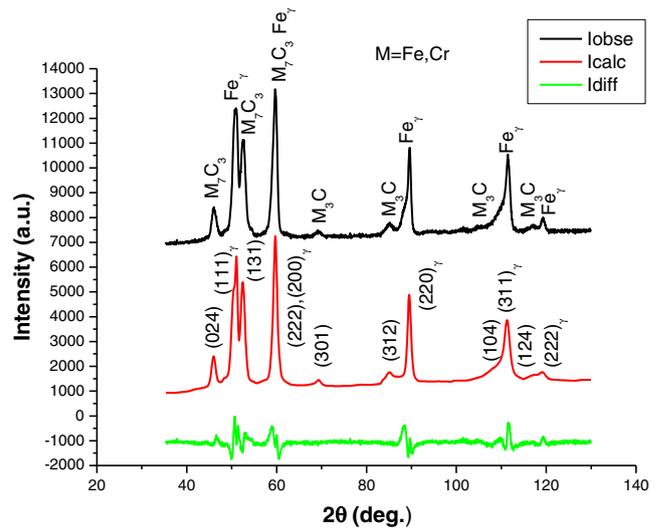
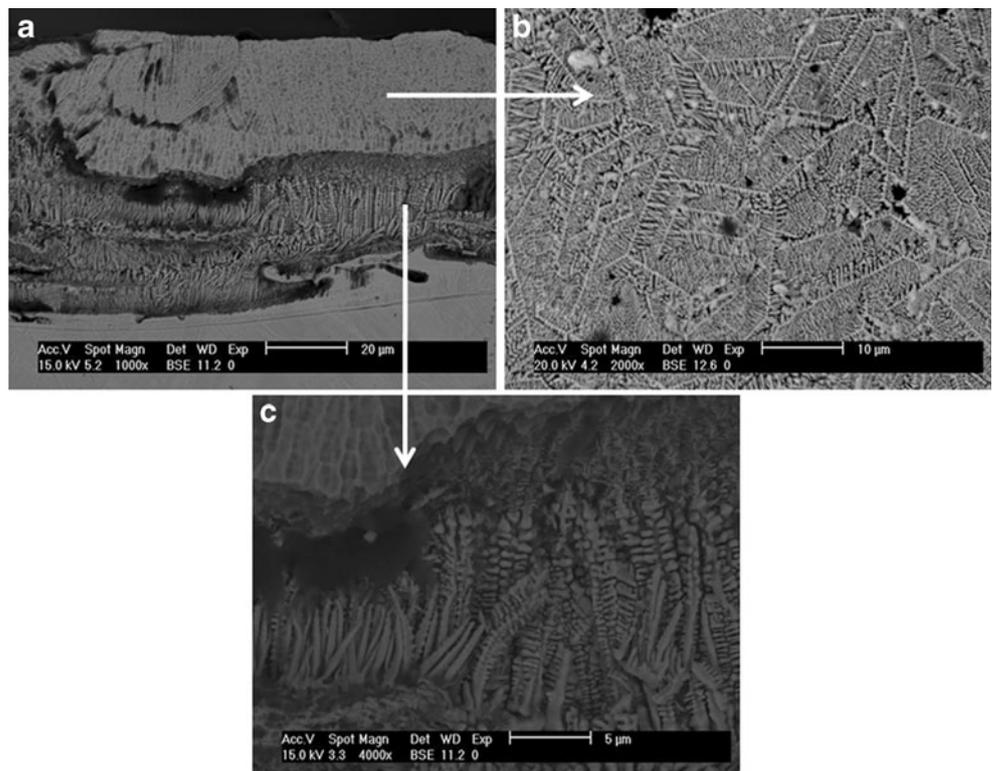


Fig. 8 X-ray spectrum of the white layer, where I_{obse} and I_{calc} are the observed and calculated intensity respectively and I_{diff} is the difference spectra

of the conventionally finished surface (PCT=38°C; E_p , 450 mV/SCE and E_r =-68 mV/SCE). This reflects that EDM deteriorates significantly the pitting corrosion resistance. Similarly, much higher pH activation values of EDM surface show a lower resistance to the pitting propagation. This weakening is more and more important with an increase of current discharge intensity.

Table 6 Lattice parameters of white layer phases calculated by Rietveld refinement method

Phase	Space group	Lattice parameters	Rietveld goodness of fit factors
γ	F m $\bar{3}$ m	Cubic $a=0.35932\pm 0.00001$ nm $\alpha=\beta=\gamma=90$	Bragg R factor, 5.88 RF factor, 2.90
M_3C , $M=Fe, Cr$	P n m a	Orthorhombic $a=0.50414\pm 0.00031$ nm $b=0.76662\pm 0.01018$ nm $c=0.44757\pm 0.00017$ nm $\alpha=\beta=\gamma=90$	Bragg R factor, 8.45 RF factor, 5.70
M_7C_3 , $M=Fe, Cr$	P n m a	Orthorhombic $a=0.44801\pm 0.00004$ nm $b=0.69279\pm 0.00007$ nm $c=1.22170\pm 0.00043$ nm $\alpha=\beta=\gamma=90$	Bragg R factor, 2.66 RF factor, 1.52

4.2.2 IGC sensitivity

The values of the I_r/I_a ratio varies from EDM finishing conditions ($I_r/I_a=28\%$) to EDM roughening conditions ($I_r/I_a=60\%$). This confirms a clear sensitivity to IGC of EDM surfaces in comparison to the conventionally machined surface ($I_r/I_a=0\%$). SEM metallographic examinations of the EDM surface before and after DL-EPR testing reveal selective dissolution in the inter-dendritic areas of re-solidified structure of the white layer (Fig. 16).

4.2.3 Susceptibility to RSCC

SEM observations of specimens, immersed during 48 h in a 40% $MgCl_2$ heated at 140°C, show secondary microcrack network that are very shallow and clearly separated from the primary EDM thermal cracks (Fig.17). This phenomenon is observed for roughening and finishing EDM conditions ($i_e=2-50$ A) and reveals the high susceptibility to RSCC of

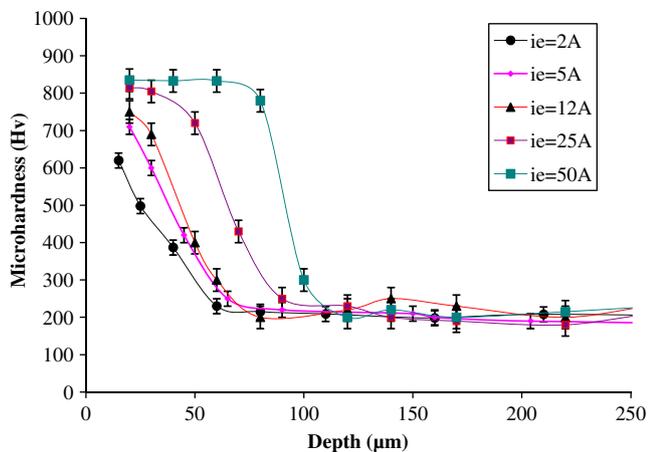


Fig. 9 White layer hardness profiles

white layer. This suggests that the threshold cracking stress in this environment is well below 200 MPa (Fig. 13).

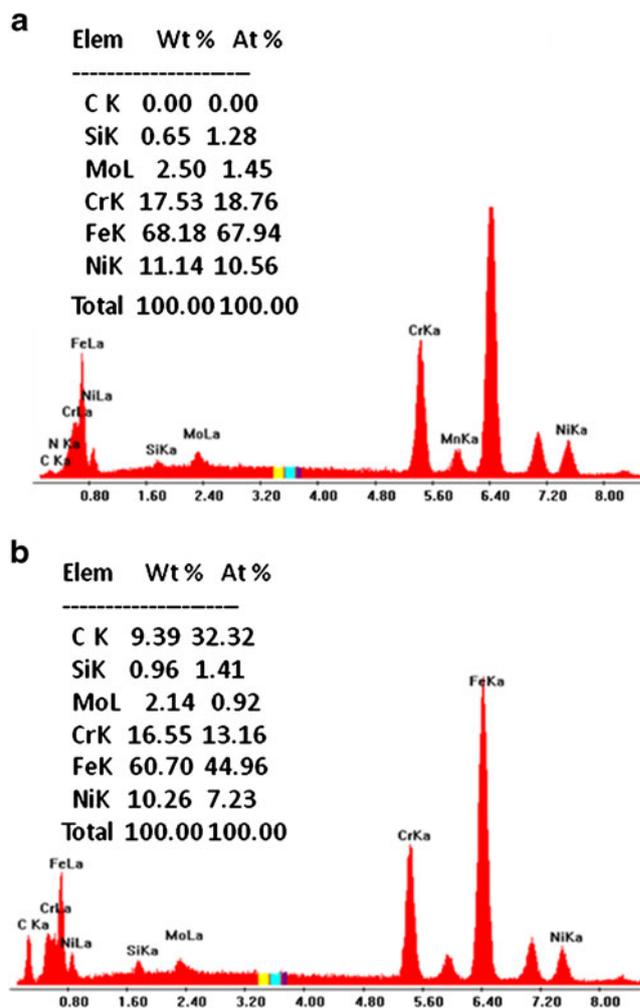


Fig. 10 EDX analysis showing carbon-enriched EDM surface; **a** bulk material carbon content and **b** white layer carbon content

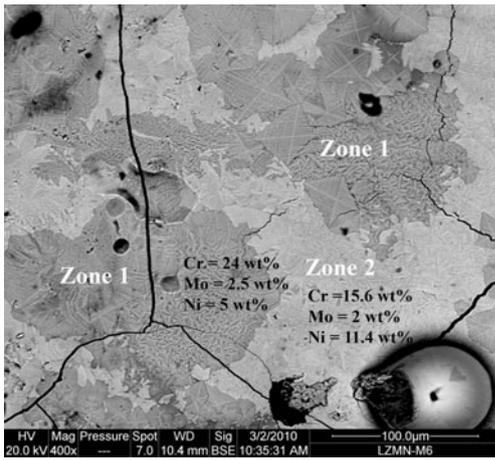
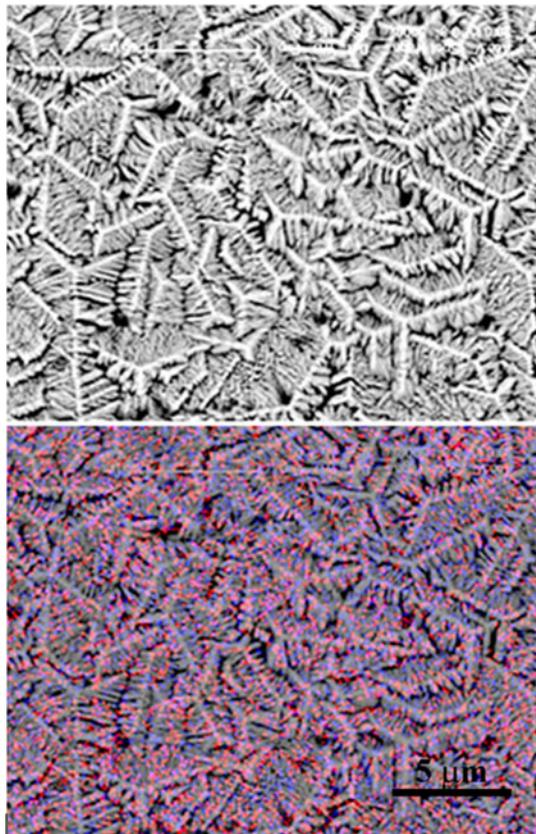


Fig. 11 Structure and chemical composition heterogeneities of EDM surface

5 Discussion

The thermal cycles created by the discharge sparks during EDM are characterised by a high heating ($V_h \approx 10^5$ °C/s) and cooling rates ($V_c \approx 10^8$ °C/s). In addition, the surface layer temperature rises to values ranging from 10^3 to $2.5 \cdot 10^3$ °C for discharge current intensity as low as 2 A, according to data from the literature [4–7]. These thermal cycles that occur in the presence of a dielectric liquid and electrode

Fig. 12 X-ray mapping showing chromium and molybdenum distribution in dendritic structure of recast layer. **a** Dendritic structure and **b** X-ray Cr-K α (purple) and Mo-K α (blue) mapping



 Molybdenum
 Chromium

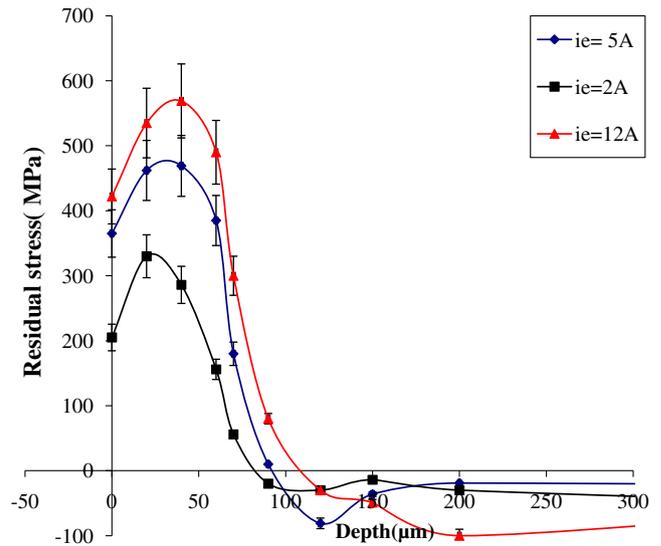


Fig. 13 Tensile residual stress distribution in the EDM affected layers

tool, promote thermal, physical and chemical exchanges. It was previously shown that these exchanges depend on the type of the electrode, workpiece material and the dielectric fluid. The commonly found is that the white layer resulting from solidification of melting zone exhibits high hardness and good adherence to bulk. However, it contains surface irregularities, multiphase materials and microcracks which

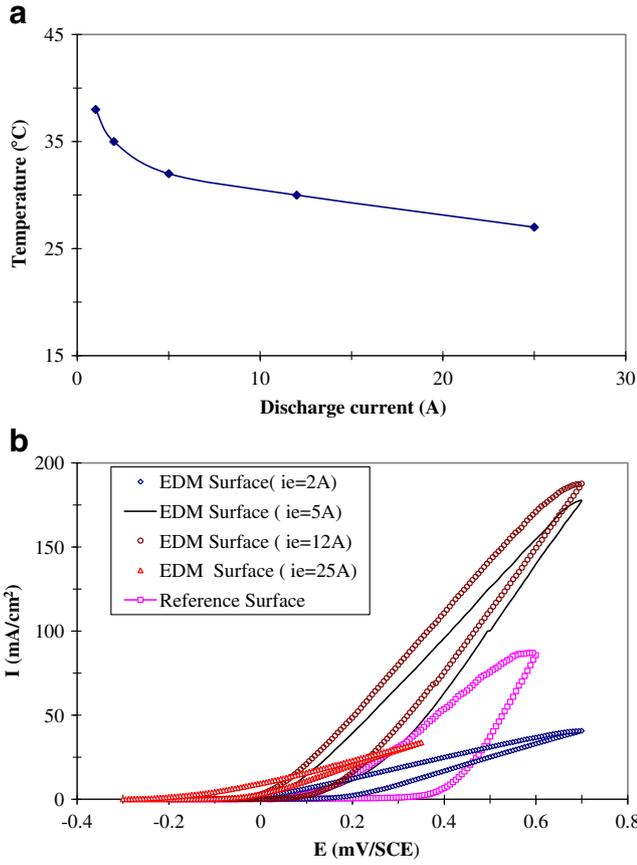


Fig. 14 Effect of EDM conditions on the white layer pitting corrosion resistance. **a** Effect of EDM conditions on the critical pitting temperature (CPT) and **b** effect of EDM conditions on the evolution of pitting (Ep) and repassivation (Er) potentials

may constitute a problem for many applications [5, 8–10, 13, 23]. The main changes that have been identified near the surface, in the case of the roughening and finishing

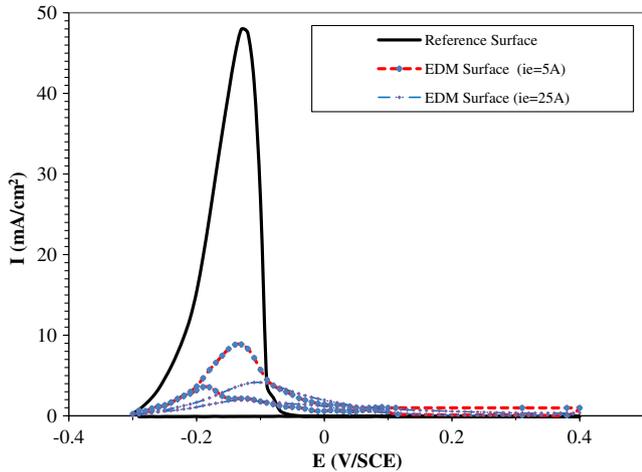


Fig. 15 Effect of EDM conditions on the white layer intergranular corrosion resistance, evaluate by DL-EPR test

Table 7 Effect of the AISI316L SS white layer properties on the corrosion resistance

Surface state	(A) Surface mode preparation		(B) Micro-geometrical change		(C) Surface changes		(D) Localised corrosion resistance									
	R_a (μm)	R_r (μm)	Micro-structural changes	Chemical changes	Mechanical changes/surface residual stress (MPa)	Surface integrity		Pitting initiation		Pitting propagation		Pitting and crevice resistance		Intergranular RSCC		
						CPT ($^{\circ}\text{C}$)	Ep (mV/ECS)	Er (mV/ECS)	pH	Sensitization to IGC	I_r/I_a (%)	RSCC cracking	RSCC cracking			
														Er (mV/ECS)	Er (mV/ECS)	Er (mV/ECS)
Reference	Polished with diamond paste	0.15	1.4	No	No	-300	No defects	38	450	-68	1.27	0	No RSCC cracking	No RSCC cracking		
EDM	$i_e=2$ A	10	30	Dendritic structure and chromium carbides precipitation	Carbon enriched austenite and interdenritic chromium depletion	+205	Thermal cracks, micro-pores and material deposits	32	230	-50	5.35	28.37	RSCC cracking	RSCC cracking		
	$i_e=5$ A	11.7	90			+365		30	87	-292	6.5	32.39				
	$i_e=12$ A	12.5	120			+422		25	45	-133	NE	29.18				
	$i_e=25$ A	19.2	220			+460		20	18	-242	NE	46.88				
	$i_e=50$ A	25.7	320			+520		16	12	-300	NE	60				

E_p pitting potential, I_{GC} intergranular corrosion, $RSCC$ residual stress corrosion cracking, E_r repassivation potential, I_r density of reactivation current, I_a density of activation current, CPT critical pitting temperature, NE non-evaluated

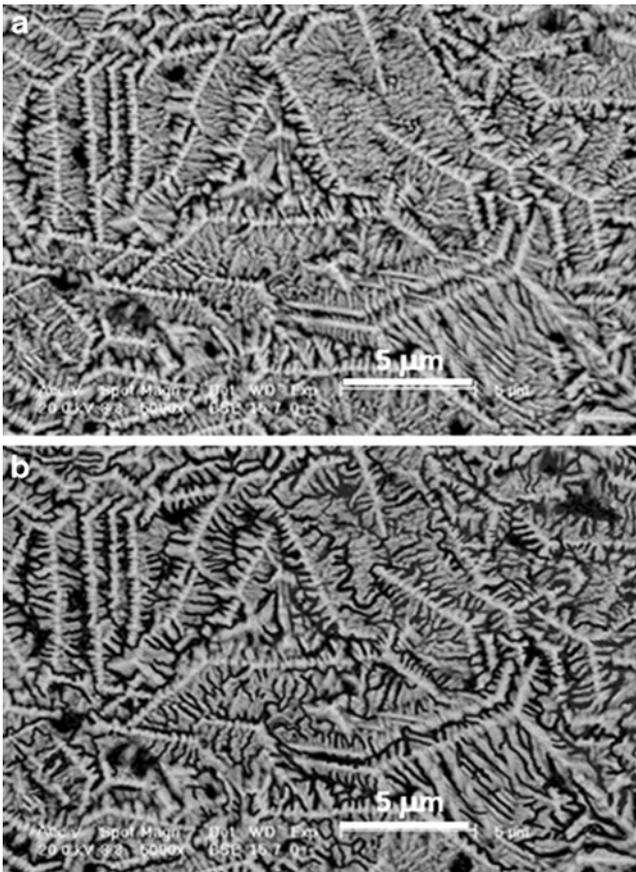


Fig. 16 Selective dissolution of interdendritic spaces after DL-EPR test performed in a solution of H_2SO_4 0.5 M + NH_4SCN 0.01 M at 20°C and under 2.5 mV/s **a** before DL-EPR test and **b** after DL-EPR test interdendritic dissolution

conditions of AISI316L SS with graphite electrode and paraffin oil as dielectric fluid, are summarised in Table 7 (columns B and C). These changes concern the white and heat-affected layers. The depth of the white layer and heat-affected layers is lower than $150\ \mu\text{m}$. The finishing mode produces a much thinner layer ($5\ \mu\text{m}$) in contrast to high energy roughening mode ($80\ \mu\text{m}$). The depth of heat affected layer is around $50\text{--}150\ \mu\text{m}$. The white layer demonstrates changes in surface texture, microstructure, composition and residual stress state that control the corrosion resistance of EDM surface.

5.1 Surface texture changes

EDM surface is characterised by pores and craters, which are characteristic of melting drop resulting to a higher roughness texture comparatively to conventional machining process. EDM R_t varies from $300\ \mu\text{m}$ for roughening conditions to $30\ \mu\text{m}$ for finishing conditions, while the turning roughness is usually less than $30\ \mu\text{m}$ and the grinding roughness is less than $10\ \mu\text{m}$.

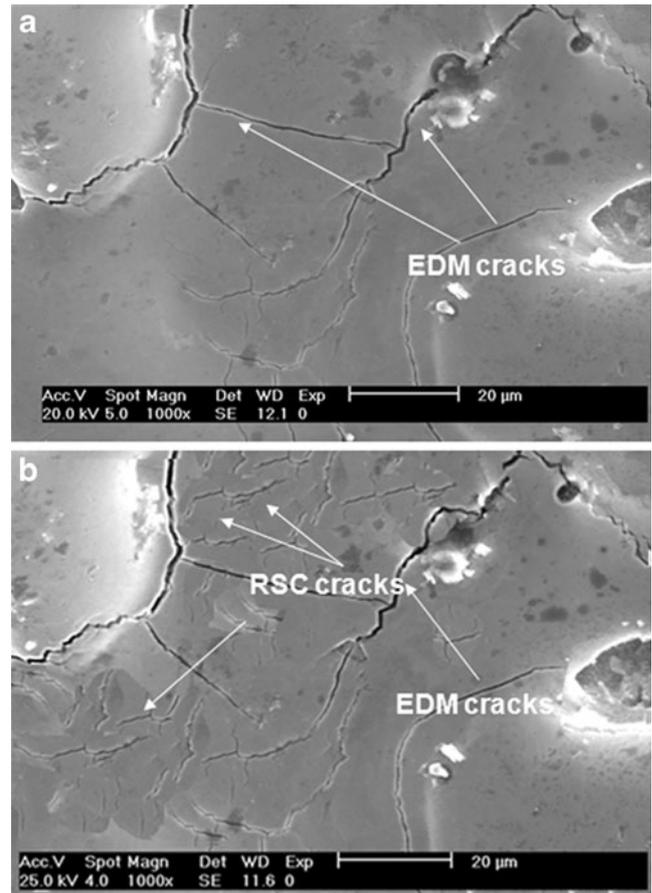


Fig. 17 RSCC of AISI316L SS EDM surface. **a** Before immersion and **b** after immersion during 48 h in 40 % MgCl_2 solution heated at 140°C

5.2 Microstructure and chemical changes

The white layer is composed by recast sublayers of some microns in thickness and melted sublayers with dendritic structure. It consists of γ , M_3C and chromium carbides precipitation (Cr_7C_3). Similar phases have been observed by Cusanelli et al. [10] in the case of EDM ferritic steel. These phases are metastable, they are formed at high temperature with an unknown mechanism. Therefore, a nanostructured morphology can be suspected regarding the high heating and cooling rates required by this process. Carbon is provided from dielectric liquid (paraffin oil) and graphite electrode that diffuse in the melt material leading to the carbide precipitation and not from diffusion of the base material as it has been demonstrated by Cusanelli [10] using a silicon oil as a dielectric fluid. As a consequence, the high carbon content of the white layer revealed by EDX analysis is mainly associated to the distribution of carbides into the recast layer. The high hardness of the white layer is correlated to the chromium and iron carbide dispersion and to the carbon enriched austenite that contributes significantly in its brittleness.

5.3 Mechanical changes and corrosion resistance

The EDM surface residual stress distributions are the consequence of surface layer contractions and induced volume changes resulting from high cooling rate. The carbide volume is larger than austenite one. The volume increase reaches 24 % for ϵ -carbide and 48 % for Cr_7C_3 . These changes involve a high tensile stress state in the austenite and that induces the crack nucleation in this phase. The cracks rapidly grow by fast cooling of the brittle white layer and lead to near-surface release of tensile residual stress. All these changes, denoting an extensive electrochemical reactivity action, take place and explain the difference in the localised corrosion behaviour in chloride environments between AISI316L SS surfaces, machined by this process, and reference surfaces polished with diamond paste.

5.3.1 Resistance to pitting initiation and growth

The decrease in pitting corrosion resistance of AISI316L SS EDM surface is mainly related to texture irregularities. Indeed, the values of the activation E_p mark significant drops compared to the value of the reference state (450 mV/SCE), for roughening EDM machining conditions ($i_e > 12$ A) while finishing conditions ($i_e < 5$ A) affect slightly the pitting and crevice corrosion resistance. This weakening of EDM surface corrosion resistance is confirmed by a decrease of corrosion pitting temperature (CPT) and an increase of pH activation and it is correlated with roughness. The increase of roughness of the resultant surface lead to pitting and crevice corrosion attack. That is why a substantial increase in the pore and crater size has been observed after corrosion testing. Such behaviour might be attributed to a mechanism involving the role of craters, pores and cracks to concentrate very reactive ions (Cl^-). The galvanic phenomena might also be generated among different phases that constitute the white layer surface by favouring pitting nucleation and growth on the chromium depleted zones. These hypotheses need to be supported by SEM corrosion morphology examination at higher magnification.

5.3.2 Resistance to IGC

The white layer is a multiphase material (γ , ϵ -carbides and Cr_7C_3) with structure and chemical heterogeneities that promote the easily soluble chromium and molybdenum depleted regions. Based on this information, a more detailed explanation can be given for IGC sensitization with the fact that Cr_7C_3 is an insoluble phase while interdendritic space dissolves in the corrosion medium. That is why the I_p/I_a ratio increases comparatively to the surface reference. The chromium depletion is the consequence of the chromium carbide precipitation in the melting structure of the white layer. The

chromium level below 13 at.% is likely reached at the interdendritic space to enhance IGC sensitisation according to Sidhom et al. [29] works on the same austenitic stainless steel. Moreover, the high values of the I_p/I_a ratio reveal the existence of relatively large areas that are heavily depleted from chromium and molybdenum [29]. The fact that these values I_p/I_a going from 30 % for finishing conditions to 60 % for roughening conditions, indicates that this phenomenon is mainly related to the existence of the white layer microstructure and chemical heterogeneities at the submicron scale. Indeed, the very fast solidification of the material melted by electric discharge, under non-equilibrium conditions, does not allow the homogenization of chromium contents. Such homogenization is required for the IGC desensitisation of the white layer.

5.3.3 Resistance to RSCC

The susceptibility to RSCC of AISI316L SS machined surfaces results from the tensile residual stresses induced by the EDM process. The threshold of cracking stress for the tests performed in this study seems to be reached for all the machining conditions under low or high discharge current, according to data on the same material machined by other processes [30, 31]. The low depth of the EDM tensile residual stress explains the RSCC short cracks.

5.3.4 Corrosion resistance improvement of EDM surface

The improvement of the EDM surface pitting and crevice corrosion needs a control of surface irregularities and microstructure and chemical heterogeneities by finishing machining conditions ($i_e < 5$ A). Moreover, an annealing treatment after EDM machining leading to chemical homogenisation and tensile residual stress release prevents the IGC and the RSCC. Indeed, annealing treatment after EDM ensure material self healing, by the elimination of chromium depleted zones (diffusion phenomenon that is thermally activated) and tensile residual stress, simultaneously. Finally, the material removal of white layer by wire brushing restores the corrosion resistance of AISI316L SS.

6 Conclusion

It is clear from the results of the present study that EDM machining compromises the corrosion resistance of AISI316L SS. Surface irregularities like pores, craters and the cracks affect substantially the pitting and crevice corrosion resistance. Structure and chemical heterogeneities of the white layer at micron and submicron scales alter the IGC resistance. The chromium-depleted zone resulting from Cr_7C_3 at high temperature and during cooling promotes

the dissolution of interdendritic space material. The tensile residual stresses resulting from rapid cooling and phase changes in the white layer induce cracking of AISI316L SS in chloride medium. Therefore, the white layer generated by the EDM process has to be annealed or completely removed for implants made in AISI316L SS to avoid the formation of toxins.

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